TWO DECADES OF ENVIRONMENTAL ISOTOPE RECORDS IN CROATIA: RECONSTRUCTION OF THE PAST AND PREDICTION OF FUTURE LEVELS

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ABSTRACT. A two-decade-long record of environmental isotopes (²H, ³H, ¹⁴C, ¹⁸O) in Croatia is presented and the data are statistically analyzed. The atmospheric ¹⁴C activity for the period before the actual measurements started is reconstructed by measuring tree rings from the clean-air sites, and the past tritium activity in precipitation is estimated by the correlation of our data with the Vienna station record. The long-term ³H record helped to determine a locally contaminated sampling site, and new clean sites are put into operation. The ¹⁴C data were fitted by an analytical function and the prediction of future levels is given assuming that the rate of the ¹⁴C releases remains constant. From the long-term stable isotope data record, the local meteoric water line and the temperature gradient of δ^{18} O in precipitation are determined.

INTRODUCTION

Monitoring of environmental levels of radioactive (14 C, 3 H) and stable isotopes (2 H, 13 C, 18 O) in the atmosphere and biosphere started about four decades ago, approximately at the same time as the significant anthropogenic disturbance of natural distributions of radiocarbon and tritium occurred. In the Radiocarbon and Tritium Laboratory of the Ruder Bošković Institute in Zagreb (Croatia) measurements of atmospheric 3 H and 14 C have been performed since 1976 and 1978, respectively. A few years later, the lab also started monitoring stable isotopes (2 H and 18 O) in precipitation samples in cooperation with the Jožef Stefan Institute in Ljubljana, Slovenia. These two stations have been continuously active since then, as shown in Table 1.

The disturbances of natural distributions of ³H and ¹⁴C made these isotopes very important tracers in, *e.g.*, environmental, climatological and hydrological studies. A global network for ²H, ³H and ¹⁸O isotopes in monthly precipitation was organized by the International Atomic Energy Agency (IAEA) and the World Meteorological Organization (WMO). The results have been regularly published by the IAEA (1969, 1970, 1971, 1973, 1975, 1979, 1983, 1986, 1990, 1994). The data from *ca.* 200 stations from around the world consist of tritium concentration and stable isotope (²H, ¹⁸O) content of monthly precipitation samples, and also relevant meteorological data: amount and type of precipitation, mean monthly surface air temperature and humidity. All available data from the network were recently analyzed by Rożanski *et al.* (1991, 1993). Several representative stations with different climatic conditions demonstrated temporal and spatial variations of isotope distributions. The importance of long-term stable isotope records for climatic studies is pointed out (Rożanski *et al.* 1993), as well as the importance of ³H monitoring because of the increasing input from anthropogenic sources (Rożanski *et al.* 1991).

The ³H and stable isotope data measured at the Zagreb and Ljubljana stations have been regularly published since 1976 and 1981, respectively (IAEA 1983, 1986, 1990, 1994). In addition to these two sites, several other sampling sites have been included in the monitoring, as shown in Table 1. All the sampling stations, except Rijeka (Fig. 1) have a continental climate. The nearest continental station of the IAEA network is Vienna–Hohe Varte, and our results are compared mostly with data from that station. The Rijeka station has a Mediterranean climate, and the ³H activity in Rijeka precipitation is compared to that in Genoa (Italy), which is the closest Mediterranean station.

Natural distribution of atmospheric ¹⁴C was disturbed during the early 1960s due to the release of bomb-produced ¹⁴C into the atmosphere. The new global distribution can be further locally affected by various nuclear facilities (nuclear power plants (NPPs), medical and research institutions that use

labeled material) or by intensive combustion of fossil fuels in metropolitan or industrial areas. While the first type of ¹⁴C sources tends to increase the local ¹⁴C activity, the latter source (fossil fuel contribution) tends to decrease the local ¹⁴C activity (the Suess effect) and is therefore considered as a "sink" of atmospheric ¹⁴C.

are given in brackets and Location and		Altitude	- .	Active sampling
site name	Coordinates	(m asl)	Isotopes	period
Zagreb-RBI* (R)	45°49' N 15°58' E	165	² H, ³ H, ¹⁸ O, ¹⁴ C	Since 1976
Zagreb-Grič (G)	45°48' N 15°57' E	158	² H, ³ H, ¹⁸ O	Since 1996
Puntijarka (Pn) Mt. Medvednica	45°55' N 15°57' E	988	² H, ³ H, ¹⁸ O, ¹⁴ C	Since 1995
Plitvice (3)	44°50′ N 15°35′ E	536	³ H, ¹⁴ C	1976–1990
Rijeka (4)	45°20′ N 14°30′ E	~20	³ H	1986–1987
Ljubljana (2)	46°04′ N 14°31′ E	299	² H, ³ H, ¹⁸ O	Since 1981
Libna (L) 1.5 km from NPPK*	45°55' N 15°31' E	~180	¹⁴ C	1984–1986
Dobova (D) 12 km E from NPPK	45°50' N 15°40' E	~170	¹⁴ C	1984–1986
Pesje (P) 3 km E from NPPK	45°54' N 15°35' E	~160	¹⁴ C	1984–1986
Bistra (B) 34 km E from NPPK	45°54' N 15°50' E	~200	¹⁴ C	1984–1986

TABLE 1. Sampling sites for environmental samples: geographical position, monitored isotopes, and the period when the sites have been active. Sampling site identification symbols are given in brackets and are as in Figure 1.

*RBI=Ruđer Bošković Institute; NPPK= Krško NPP

For atmospheric ¹⁴C, no such large global network, such as that for isotopes in precipitation has been established. Several stations in Germany (Levin *et al.* 1980, 1992, 1995; Levin and Kromer 1997; Segl *et al.* 1983), Canada (McNeely 1994), Norway and the Netherlands (Meijer *et al.* 1994) have long records of ¹⁴CO₂ activity. The University of Heidelberg organized a network of seven stations distributed world-wide between 82°N and 71°S (Levin *et al.* 1992, Levin and Kromer 1997) for monitoring of atmospheric CO₂ in the troposphere. In addition, isotopic composition of carbon in atmospheric CO₂ has been measured for more than a decade in Kraków, Poland (Kuc and Zimnoch 1998) and in Sweden (Olsson 1989).

Partial presentation of our data can be found in previous papers (Horvatinčić *et al.* 1986, 1992, 1996a, 1996b; Krajcar Bronić *et al.* 1986, 1992; Obelić *et al.* 1986, 1987, 1992; Kozák *et al.* 1989). We present here the complete isotope data record (²H, ³H, ¹⁴C, ¹⁸O) of atmospheric samples measured during the last two decades in our laboratory. Our statistical analysis of the stable isotope and

³H data in precipitation is compared with the analysis of Rožanski *et al.* (1991, 1993). ¹⁴C activity of atmospheric CO₂ is compared with other atmospheric measurements (Levin *et al.* 1995; Levin and Kromer 1997; McNeely 1994; Meijer *et al.* 1994).

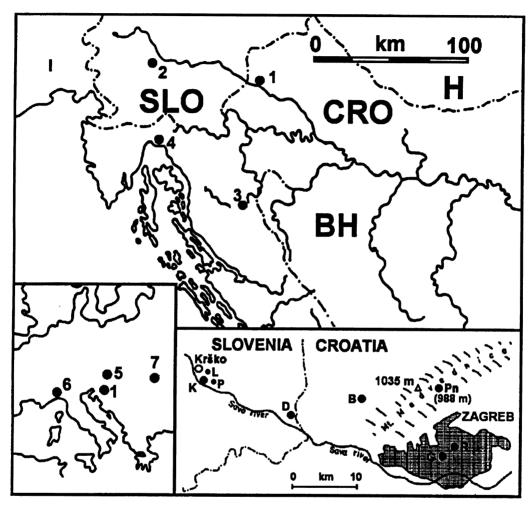


Fig. 1. Sites for environmental samples in Croatia and Slovenia. 1. Zagreb; 2. Ljubljana; 3. Plitvice; 4. Rijeka. Lower left corner: the European stations, near our sampling sites. 5. Vienna (Austria); 6. Genoa (Italy); 7. Mt. Matra (Hungary). Lower right corner: locations in the city of Zagreb, at Mt. Medvednica, and around the Krško NPP. K=NPPK; L= Libna; P=Pesje; D=Dobova; B=Bistra; Pn=Puntijarka, G=Zagreb-Grič; R=Zagreb-RBI.

Sample Preparation and Measurement Methods

Experimental methods of sample collection, preparation and measurement used in our laboratory have been previously discussed in detail (Horvatinčić 1980; Horvatinčić *et al.* 1986, 1989, 1990; Obelić *et al.* 1986). We describe here only the most important details.

Organic samples for ¹⁴C activity measurement are chemically pretreated by the HCl-NaOH-HCl (or A-B-A) method (Srdoč *et al.* 1971; Krajcar Bronić *et al.* 1995) and then combusted in a stream of purified oxygen. The preparation of cellulose from separated tree rings has been described in detail

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by Obelić *et al.* (1992), Kozák and Rank (1981) and Kozák *et al.* (1989). Samples of atmospheric CO_2 are collected by exposing *ca.* 100 mL of saturated carbonate-free sodium hydroxide solution in a tray protected by a wire-mesh to the open atmosphere. The exposure time is typically one month. The absorbed CO_2 is released in the laboratory by dissolving the sodium carbonate in diluted HCl. The CO_2 obtained either by combustion of organic samples or by dissolution of carbonates by HCl is then catalytically converted to CH_4 by hydrogenation over a Ru catalyst. Purified methane is used as a filling gas in a proportional counter for ¹⁴C activity measurement. For ³H measurements, a wallless proportional counter filled with methane is used. Water samples are converted to methane by reaction with aluminum carbide at 150°C (Horvatinčić 1980).

Air moisture (atmospheric water vapor) has been sampled regularly since 1988, although some occasional samplings were done before that time. All the samples of atmospheric moisture at the Institute and in the various places in the city are collected by absorption of air moisture on silicagel in a flow mode (Horvatinčić *et al.* 1989; Bistrović *et al.* 1994). Evaporation and isotope-exchange with ³H-free water is used for regular checks of laboratory air, while the method of freezing out of the water vapor on Al plates cooled with liquid nitrogen is used for fast (immediate) sampling when necessary. Comparison of different sampling methods gives satisfactory agreement, taking into account different time periods needed for the sample collection.

Stable isotopes ¹⁸O and ²H in water samples are measured by the mass spectrometer MAT 250 at the Jožef Stefan Institute in Ljubljana (Slovenia). Water samples for δ^2 H determination are reduced to H₂ by passing over hot Zn, whereas for ¹⁸O measurement, water samples are equilibrated with CO₂ gas of known isotopic composition.

Our laboratory participated in several intercomparison studies for both ¹⁴C and ³H isotopes (Scott *et al.* 1990, 1992; Rožanski *et al.* 1992). The results of ¹⁴C intercomparisons have been discussed in Horvatinčić *et al.* (1990) and Krajcar Bronić *et al.* (1995). The agreement between our results and the median values reported for each intercomparison sample was very good. Intercomparison of ³H activities in water samples, organized by the IAEA in 1986 and 1994, also showed good agreement between our results and the median values (Hut 1986).

The results of isotope content are reported throughout the paper in the following units.

- ¹⁴C as Δ¹⁴C (in ‰) deviation from the standard sample, 95% activity of the NBS Oxalic acid (I) (Stuiver and Polach 1977),
- ³H activity in Bq L⁻¹, which is related to the commonly used TU (³H unit) as 1 TU = 0.118 Bq L⁻¹,
- Stable isotope content is expressed in δ-notation, as relative deviation from the isotope content of the standard SMOW (standard mean ocean water) sample.

RESULTS

Radiocarbon

Our monitoring of ¹⁴C includes three different types of sites (Table 1, Fig. 1):

 Sites far from large cities and nuclear power facilities, which are supposed to reflect only the global disturbance of the atmospheric ¹⁴C activity—the so-called "clean-air" sites (Plitvice, Puntijarka). After 1990, the "clean-air" site at Plitvice was not reachable, hence we started our monitoring at the Puntijarka station, on Mt. Medvednica near Zagreb. The station is 15 km from the city of Zagreb and 988 m above sea level (asl), and prevailing winds blow from Puntijarka to Zagreb.

- 2. A densely populated industrial center (Zagreb), where the Suess effect may be expected. However, there are also large research and medical institutions that use radioactive-labeled material, and some local contamination may be observed.
- 3. Several locations around the Krško NPP, which is situated 37 km northwest of Zagreb in the Republic of Slovenia. Libna is the closest monitoring station, only 1.5 km from the plant smokestack. Other sampling sites (Pesje, Dobova, Bistra) were chosen according to their geography and the prevailing weather conditions.

We monitored ¹⁴C in the biosphere through atmospheric CO₂, or by annual or short-lived plants (seasonal fruits or vegetable, grass) and leaves. All these materials closely monitor the atmospheric ¹⁴C concentration (Levin *et al.* 1980; Otlet *et al.* 1992). We collected organic material and atmospheric CO₂ at "clean-air" sites Plitvice and Puntijarka (Table 1, Fig. 1), and at various sites in the vicinity of the Krško NPP, while in the vicinity of Zagreb, only atmospheric CO₂ samples were collected. We also measured ¹⁴C activity in tree rings from "clean-air" sites Plitvice and Matra, and from the vicinity of the NPP. The results obtained around the Krško NPP were discussed extensively by Obelić *et al.* (1986, 1987).

In Figure 2, we present the measured ¹⁴C activity in short-lived plants collected at the Plitvice site and in atmospheric CO₂ at the Zagreb-RBI site for the period 1979–1996. For comparison, we also show ¹⁴C activity in rings (1979–1983) of a linden tree that was felled at the Libna station. A continuous decrease in the ¹⁴C activity at all three sites is evident. Segl *et al.* (1983) determined for the period 1976–1982 an average decrease in Δ^{14} C of 20% per year for "clean-air" continental stations.

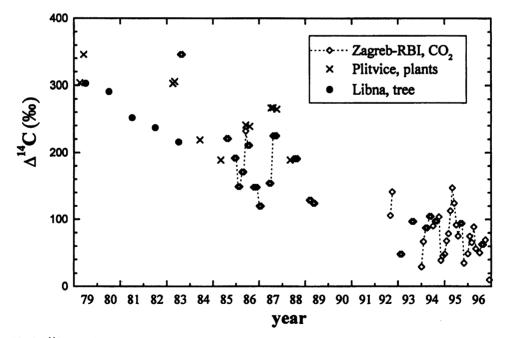


Fig. 2. Δ^{14} C values in monthly samples of atmospheric CO₂ collected at Zagreb-RBI and in some short-lived plants collected at the Plitvice site 1979–1996. ¹⁴C activities taken from rings from the linden tree (Libna station) are also shown. The broken line connecting subsequent data points (Zagreb-RBI, CO₂) indicates seasonal changes in Δ^{14} C.

The decrease of ~22‰ per year is obtained for the period 1979–1983 from our tree-ring data (Fig. 2). However, in recent years, the yearly decrease in Δ^{14} C is lower: for the period 1984–1996 the Δ^{14} C data measured at the Zagreb-RBI station show a decrease of 12‰ per year, and the data measured at Plitvice in the period 1979–1990 a decrease of 10.6‰ per year. Similarly, Levin *et al.* (1992) report a decrease of 14.1‰ and 9.7‰ per year for the periods 1983–1985 and 1985–1989, respectively, at the Schauinsland station.

As Figure 2 shows, seasonal fluctuations in atmospheric Δ^{14} C are superposed to the mean yearly values: the highest ¹⁴C activity is measured during summer, and the lowest in winter (see also Fig. 3). The analysis of the deviations of the monthly values from the mean yearly Δ^{14} C for four full-year cycles (1986, 1994-1996) shows that the averaged maximal deviations are observed in May (41‰ above the mean Δ^{14} C) and in December (40‰ below the mean yearly Δ^{14} C), giving average yearly peak-to-peak variations of 81‰. Seasonal peak-to-peak variations at the "clean-air" site Shauinsland were determined to be 40‰ (Segl et al. 1983), while the seasonal peak-to-peak variations at the Kraków site, which is influenced by the fossil-fuel carbon, were 56‰ (Kuc and Zimnoch 1998). Thus, higher variation in Zagreb indicate greater influence of the fossil-fuel carbon (i.e., the stronger Suess effect) during the winter months. In Figure 3, we compare Δ^{14} C values measured in the city of Zagreb (Zagreb-RBI site) with those measured at the clean-air site Puntijarka during 1995 and 1996. In the colder time of the year, the Δ^{14} C values in the city of Zagreb are much lower than at Puntijarka, confirming the conclusion derived from the comparison with the Schauinsland data. The mean yearly Δ^{14} C values in Zagreb (81‰, 88‰ and 56‰ for 1994, 1995 and 1996, respectively) are lower than those measured at Schauinsland (117%, 111% and 102%, respectively; Levin and Kromer 1997) or in Kraków (120-125‰; Kuc and Zimnoch 1998) for the same period. Even the mean Δ^{14} C at Puntijarka (82.4‰ for 1996) is lower than the above-mentioned values, although this site does not seem to be influenced by fossil-fuel carbon.

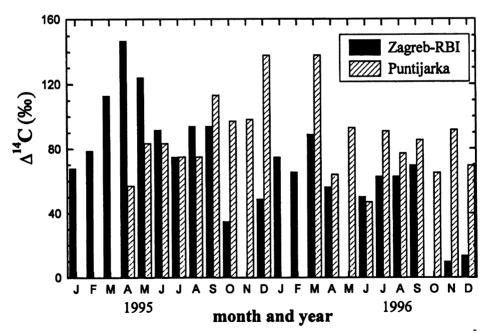


Fig. 3. Comparison of the Δ^{14} C values in the city of Zagreb and at the clean-air station Puntijarka for 1995–1996

A detailed inspection of the atmospheric ¹⁴C activity around the Krško NPP results in slight differences among various sampling sites. The average Δ^{14} C values in atmospheric CO₂ measured for the period 1984–1986 at stations Zagreb, Dobova and Bistra (Fig. 1) were (208 ± 12)‰, (207 ± 11)‰ and (216 ± 9)‰, respectively. The differences between these sites are not significant, according to the given uncertainties. Only the samples from Libna station, closest to the Krško NPP, gave somewhat higher mean Δ^{14} C values, (230 ± 19)‰ for 1984 and (216 ± 20)‰ for 1985, than those taken at the above-mentioned locations. The highest increase was observed in July 1984 (Δ^{14} C = 270‰) and coincided with the yearly exchange of one-third of the fuel elements in the reactor core. For 1984 a mean ¹⁴C excess of 22‰ above the "clean air" level was measured at the Libna station (Obelić *et al.* 1986).

The ¹⁴C activity of tree rings from Plitvice and Mt. Matra, Hungary, is used to reconstruct the atmospheric ¹⁴C activity for the period 1950–1986. The results are presented in Figure 4 and are compared with the long-term measurements of atmospheric CO₂ (from Germany, Canada, the Netherlands), as well as with the mean Δ^{14} C values for atmospheric CO₂ at the Zagreb-RBI station. The two sets of Δ^{14} C values in tree rings agree well with the mean yearly Δ^{14} C for atmospheric CO₂ at the Schauinsland station. All the curves follow each other very well; all give a maximal ¹⁴C atmospheric activity in 1963–1964 of ~900‰, and show a discontinuity in decrease in 1971–1972. However, Δ^{14} C values in Canada (McNeely 1994) are systematically higher, and the Groningen data (Meijer *et al.* 1994) are systematically lower than the clean-air site Schauinsland (Levin and Kromer 1997). Our data at Zagreb-RBI agree well with the Groningen data, and are thus lower than the Schauinsland data.

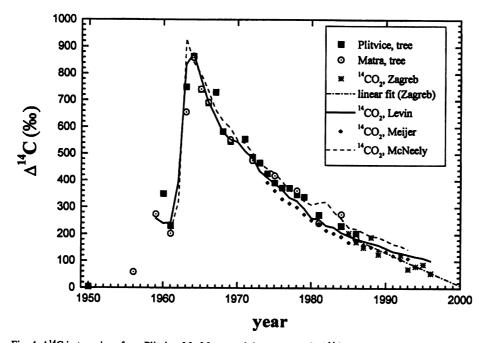


Fig. 4. Δ^{14} C in tree rings from Plitvice, Mt. Matra, and the mean yearly Δ^{14} C values in the Zagreb-RBI CO₂ samples. Our data are compared with atmospheric ¹⁴C activities at Schauinsland (Levin *et al.* 1995; Levin and Kromer 1997), Groningen (Meijer *et al.* 1994) and Canada (McNeely 1994). The dashed-dotted line represents linear fitting to the Zagreb-RBI data, and is extrapolated to the year 2000.

We fitted an exponential decay function to the Plitvice tree-ring data for 1964–1986, to the Matra treering data (1964–1986), and to the atmospheric CO₂ data by Levin *et al.* (for 1964–1996) and by McNeely (for 1963-1993). (The data for Meijer et al. (1994) could not be fitted with the same function). All the fits gave a time constant for the decay of ~16 yr. Extrapolation to the year 2000 resulted in values of ~120‰, except the Schauinsland data (Levin and Kromer 1997), which resulted in predicted Δ^{14} C of 90‰. However, recent mean atmospheric ¹⁴C activity measured in the Zagreb area is already lower than these values. As Figure 4 shows, the exponential decay curve may be divided to several time periods, and each of the period may be approximated by a linear function: 1) a sudden increase in atmospheric ¹⁴C activity for the period ~1955-1964; 2) for 1964-1969, a period of fast decrease in Δ^{14} C at ~50–65‰ per year; 3) for 1970–1980, a lower decrease rate of ~20–30‰ per year; and 4) for 1980-present, a decay rate of ~12‰ per year. Data from recent years (starting from 1982 to the most recent) are thus fitted to a linear function. The Zagreb-RBI mean yearly ¹⁴C activities gave a decrease of 12.2‰ per year, data of Meijer et al. (1994), 9.8‰ per year, and Levin and Kromer (1997), ~9‰ per year. By using the obtained linear regressions, the Δ^{14} C values can be extrapolated to future years. Our data (Zagreb-RBI) predict that due to local contamination (fossil fuel CO₂), a mean Δ^{14} C of 0‰ might be reached by the year 2000. The Groningen data (Meijer et al. 1994) and the Schauinsland data (Levin and Kromer 1997) predict for the same year slightly higher values of ~35‰ and ~62‰, respectively, and the Δ^{14} C value equal to 0‰ for the years 2004 and 2007, respectively.

Tritium

Figure 5 shows the complete record of ³H activity of the monthly precipitation of Zagreb and Ljubljana. The long-term record enabled us to recognize some local contamination in Ljubljana during 1985 and early 1986 (not shown in Fig. 5). By the end of 1993, the first sign of possible local contamination at the site Zagreb-RBI was observed. A rain gauge has been situated at the Ruder Bošković Institute since the beginning of the monitoring in 1976, and at the same location ³H in atmospheric

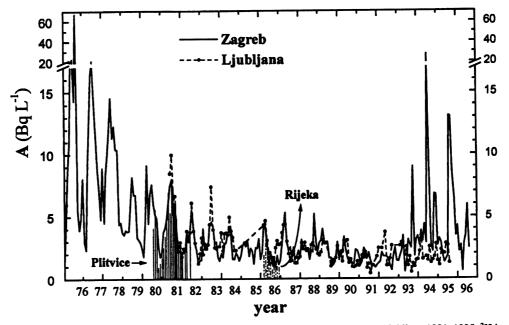


Fig. 5. A complete record of ³H in precipitation at Zagreb-RBI for 1976–1996, and at Ljubljana 1981–1995. ³H in precipitation at stations Rijeka and Plitvice is also shown. The contaminated values from the Ljubljana station in the period 1984–1985 are not shown, while the ³H activity measured at Zagreb-RBI in 1993–1996 is shown. See Figure 6 for noncontaminated values.

water vapor has been measured since 1988 (Horvatinčić et al. 1992). ³H concentration in atmospheric water at the Institute ranged between 0.07 and 0.50 Bg m^{-3} . These values were constantly higher than those at other locations $(0.02-0.04 \text{ Bg m}^{-3})$ due to local contamination with ³H, but the precipitation samples were not influenced. However, a much higher ³H concentration of 2 Bg m⁻³ was observed in summer 1992 and constantly higher values have been measured since 1994, reaching even 4.4 Bq m⁻³ in November 1994 (Bistrović et al. 1994). This long, continuous contamination is reflected also in precipitation (Fig. 5). Therefore, we had to find a new location for precipitation sampling. In 1995, we started collecting samples at Puntijarka, on Mt. Medvednica, 988 m asl and 15 km north of the city, and in 1996 also at the Zagreb-Grič site at the Hydrometeorologic Institute, in the center of Zagreb (Fig. 1). In Figure 6 we compare the ³H activity at all three sites in the Zagreb area for the period 1995–1996. ³H activities in precipitation at stations Grič and Puntijarka are almost identical, while those in precipitation collected at the RBI site are several times higher. Maximal difference was measured in December 1995–January 1996 (factor of 30). The mean ³H activity at the Zagreb-RBI site in 1995 and 1996 is 4.1 and 4.7 Bq L⁻¹, respectively, significantly higher than that at Puntijarka (1.43 and 1.35 Bq L⁻¹ in 1995 and 1996, respectively) or at Grič (1.23 Bq L⁻¹ in 1996). Thus, it is evident that our old sampling site has been locally contaminated by ³H since 1994. In further analyses, the Zagreb-RBI data up to 1993 will be used, and for 1995 and 1996, data from Puntijarka will be taken as representative for Zagreb.

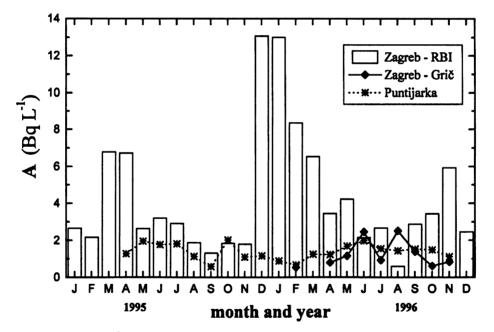


Fig. 6. A comparison of ³H in precipitation from three locations (for 1995–1996) in the Zagreb area: Zagreb-RBI, which is locally contaminated, and "clean" stations Zagreb-Grič and Puntijarka.

Characteristics of the long-term trend of ³H in precipitation at both stations Zagreb and Ljubljana (Fig. 5) are typical for the Northern Hemisphere: a seasonal structure is superposed to the basic decrease, maximal activities are measured in early summer, and the minimal winter ³H activities approached in the last years the natural, pre-bomb, level. Seasonal structure can be described as follows: In the period 1976–1992 the maximal ³H activity at station Zagreb is observed between May–July, mostly in June. A secondary maximum (Rożanski *et al.* 1991) is also observed three times in

January and February. Minimal ³H activities were almost uniformly distributed from October to February, with a slightly more frequent occurrence in December. The seasonal distribution at the Ljubljana station is characterized by the most frequent appearance of the maximum in July, no secondary winter maximum is observed, and the minimum is most frequently measured in November, although it is distributed from October to March.

For shorter periods of several years, the ³H content in monthly precipitation at two other locations in Croatia (Rijeka, Plitvice) was also recorded and is shown in Figure 5. ³H concentration in precipitation at Plitvice (typical continental climate conditions) is well correlated with that at Zagreb and Ljubljana, and seasonal variations are similar to those at other continental stations from the temperate zone of the Northern Hemisphere (Rożanski *et al.* 1991). ³H concentration in precipitation at the Adriatic coast (Rijeka) is, on the average, half of that in Zagreb. It is closely correlated with that in Genoa, while the correlation with the Zagreb station is poor.

In Figure 7, we show the mean yearly values, maximal and minimal ³H activity during a year, and the ratio maximum/minimum for Zagreb precipitation. The ratio for noncontaminated sites ranges normally from 2.2 to 5.7, without a significant trend. For Ljubljana the ratio maximum/minimum ranges also from 2.3 to 5.5. The higher ratio in 1977 (-9) of Zagreb-RBI may be a consequence of some thermonuclear tests, which also caused transient increase of the ³H level at other stations of the IAEA/WMO network (Rożanski *et al.* 1991). Higher ratios were obtained also in the period 1994–1996, and they are attributed to local contamination at the Institute area. For non-contaminated stations, at Puntijarka the ratio was 3.4 in 1995 and 2.9 in 1996, and at station Zagreb-Grič the ratio was 4.6 in 1996.

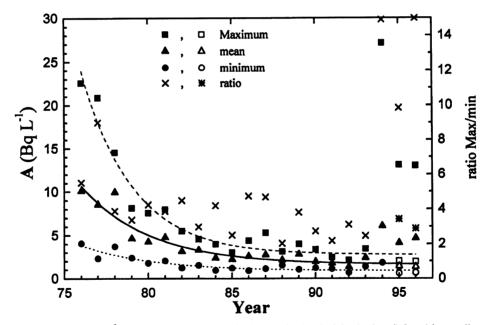


Fig. 7. The mean yearly ³H activities in Zagreb precipitation, maximal and minimal values (left scale), as well as the ratio of maximum/minimum yearly ³H activity (right scale). Full symbols represent the mean, maximum and minimum values at Zagreb-RBI, and open symbols the corresponding values at Puntijarka, while crosses and stars stand for the maximum and minimum ratios, respectively. Local contamination in the last few years caused a significant increase in the ratio for the Zagreb-RBI site, while the ratio at other two sites in the Zagreb area remained within the expected ranges. Full, broken and dotted lines represent the fits of the exponential decay function to the mean, maximal and minimal values, respectively. The Zagreb-RBI data for period 1976–1993 and the Puntijarka data for period 1995–1996 are used for the fits.

The decrease in the mean, maximal and minimal ³H activities in precipitation is very pronounced in the first decade of our record, while in the second decade all the values remain approximately constant (Fig. 7). A slight increase in 1986 and 1987 is observed both in Zagreb and Ljubljana, and also in Groningen precipitation (Meijer *et al.* 1994). An exponential decay function is fitted to the three sets of data, and the fitted curves are also shown in Figure 7. All the fitted curves support rather constant ³H activity in precipitation during the last decade. Assuming that no additional ³H will be released into the atmosphere in the future, we can predict approximate levels of ³H in future precipitation: Mean values of \sim 1.6 Bq L⁻¹ are expected, and the ratio of expected maximal (2.7 Bq L⁻¹) and minimal (0.9 Bq L⁻¹) values is 3. Any significant deviations from the expected values would then indicate some new ³H releases from fusion facilities, medical and/or industrial facilities. Monitoring of ³H activity in atmospheric water vapor, which is much more sensitive to local ³H contamination than precipitation, may also help detecting local contamination.

Reconstruction of Past Tritium Activity

As a consequence of the disturbance in the equilibrium of ³H atmospheric activity, ³H may be used as a natural tracer in various fields, such as isotope hydrology, hydrometeorology, and environmental protection studies. Such investigations require the knowledge of ³H activity in precipitation for the last 40–50 yr. ³H in precipitation was not measured in Zagreb until 1976, so the concentrations between 1953 and 1976 had to be reconstructed.

For reconstruction of the past ³H activity, we used the long ³H record at the Vienna–Hohe Varte station, as this is the nearest station with climatic conditions similar to those in our data. We found a good correlation between the yearly mean of ³H activity in precipitation in Zagreb (A_Z) and that in Vienna (A_V) for the period 1976–1994:

$$A_Z = 0.8 \times A_V + 0.4 \text{ Bq } \text{L}^{-1}, \ r = 0.985$$
 (1)

where r is the correlation coefficient. By using this relation and the A_V values for the period before 1976, the mean ³H activity in Zagreb precipitation for the period 1953–1975 is estimated.

The analysis of ³H activity in old wines of known vintage can also be a useful tool for reconstruction of the past ³H activity in a certain area. We measured ³H in several wine samples produced in north-western Croatia in the period 1974–1986. The ³H activity in three wine samples from the 1976–1986 period is well correlated with the measured mean ³H activity in precipitation, and that of the two older wine samples with the ³H activity estimated by using Equation (1) (Obelic *et al.* 1992).

We also measured the ³H activity of the water produced by combustion of cellulose prepared from separated tree rings. ³H activities in tree rings agree more or less with those in precipitation (estimated by using Eq. 1) for the most intensive bomb-test period, 1963–1967. However, the results for tree rings from the period 1975–1986 show remarkably high activities and practically no decrease in the last 15 yr (Obelić *et al.* 1992). The reason for such a high ³H activity measured in modern tree rings may be contamination introduced by the chemical procedure and chemicals used for cellulose separation. This assumption was checked by measuring the ³H activity of the water from a tree-ring sample from the period before 1950. ³H should have not been present in this sample, but we measured an activity equal to 37 ± 1 Bq L⁻¹. Such contamination is relatively small when the water from the tree-ring samples from the period 1963–1967 is concerned, but becomes more and more significant for more recent samples. Therefore, these data have not been used for the reconstruction of the past ³H activity.

Measurements After the Chernobyl Accident

The long-term records provide the data base against which the enhanced levels of radioactive isotopes can be compared. Measurements taken in April and May 1986 (after the Chernobyl accident) can illustrate the applications of such a long-term record.

After the accident at the Chernobyl NPP on April 26, 1986, the meteorological situation was such that air masses took the radioactive material towards the Baltic and the Scandinavian Peninsula during the first two days. Low pressure formed in the Genoa Bay on April 28, 1986 and caused a change in the wind direction, bringing radioactive material to central and southern Europe. Croatia was under the influence of a cold front on April 30, and the highest radioactive contamination was expected on May 1 and 2, 1986. After the first news about the accident was released, we began to collect daily samples of precipitation at three stations (Zagreb-RBI, and two locations in the Plitvice area).

Daily precipitation samples from April 29 and 30, 1986 showed no increased activity as compared with the monthly precipitation before the accident. There was no rain on May 1, and on May 2, ³H concentrations about four times higher than normal were measured at all three sampling stations. In the following days, ³H activity in the daily rain samples slowly decreased, reaching on May 8 the normal activity in spring time. The ³H of the integrated precipitation sample collected from April 30 to May 15 was 5.6 Bq L⁻¹, twice as much as expected for May. In comparison, in the samples collected in May 1985 and May 1987, the measured ³H activity was 3.0 ± 0.2 Bq L⁻¹, and 4.1 ± 0.2 Bq L⁻¹, respectively. For the samples collected on May 16–31, 1986, ³H activity was again "normal": 3.22 ± 0.23 Bq L⁻¹.

The influence of radioactive air masses containing higher ³H activity was very short-lived over northwestern Croatia, and the amount of precipitation was not abundant, leading to the conclusion that contamination of water in aquifers should be negligible. Indeed, in tapwater collected in Zagreb, as well as in several wells used for the local water supply in the Plitvice area, no increase in ³H activity was observed.

Stable Isotopes ²H and ¹⁸O in Precipitation

In this section, we analyze the long-term stable isotope data record at the Zagreb and Ljubljana stations. The average yearly amount of precipitation is 840 L m⁻² and 1474 L m⁻², and the average yearly temperature is 11.0°C and 9.9°C, in Zagreb and Ljubljana, respectively. The nearest continental station from the IAEA-WMO network is Vienna–Hohe Varte, (909 L m⁻², 10.7°C), and our results are compared mostly with this station.

In Figure 8, we present for the Zagreb station long-term monthly values for temperature (t_m) , amount of precipitation (P_m) , and stable isotope contents $\delta^2 H_m$ and $\delta^{18}O_m$ averaged over 20 yr, 1976–1996. (We denote the long-term monthly mean values with a subscript "m".) Precipitation values are rather uniformly distributed over the whole year with no seasonal maximum, while the mean monthly values of temperature and stable isotope contents show distinct seasonal behavior. Therefore, no correlation between the amount of precipitation and stable isotope data is obtained. The same conclusion is valid also for the Ljubljana precipitation samples, and therefore only figures representing data from Zagreb are shown in this section. Such a situation is typical for continental stations at the northern hemisphere (Rożanski *et al.* 1993). Maximal temperature, as well as the highest $\delta^2 H_m$ and $\delta^{18}O_m$ values, are obtained in July, and the lowest in January (in February at Ljubljana).

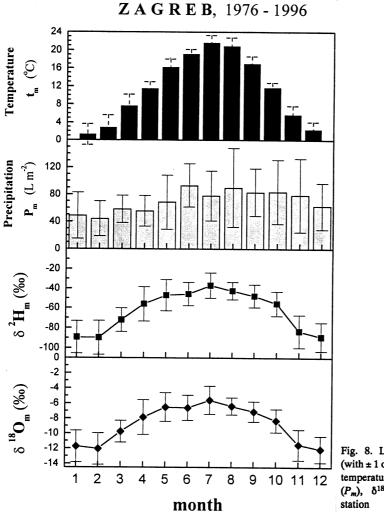


Fig. 8. Long-term monthly mean values (with $\pm 1 \sigma$, where σ is standard deviation) of temperature (t_m), amount of precipitation (P_m), $\delta^{18}O_m$ and δ^2H_m at the Zagreb-RBI station

Such good seasonality observed in both stable isotopes and temperature enables us to perform a linear regression analysis of δ^{18} O vs. the mean monthly local surface air temperatures (Fig. 9). When all the monthly data are included in the regression analysis, the following correlation is obtained

$$\delta^{18}O = (0.325 \pm 0.016) t - (12.62 \pm 0.22)\%, r = 0.83, n = 183,$$
(2)

where t is the surface air temperature, r is the correlation coefficient, and n is the number of data pairs. For the long-term mean monthly values, the following equation is obtained

$$\delta^{18}O_m = (0.327 \pm 0.024) t_m - (12.55 \pm 0.32)\%, r = 0.975, n = 12.$$
 (3)

The slope and the intercept are in both equations equal within given errors. Both lines are shown in Figure 9. For the Ljubljana station, the corresponding equations are

$$\delta^{18}O = (0.30 \pm 0.02) t - (12.00 \pm 0.26)\%, r = 0.78, n = 138,$$
(4)

$$\delta^{18}O_m = (0.295 \pm 0.021) t_m - (12.1 \pm 0.3)\%, r = 0.975, n = 12.$$
 (5)

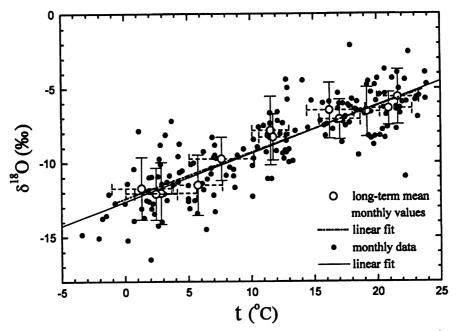


Fig. 9. Correlation between the δ^{18} O in precipitation and the local surface air temperature *t* at station Zagreb-RBI. Full symbols represent the monthly data, and open symbols with error bars are the long-term mean monthly values (t_m and δ^{18} O_m). Linear fits to all data and to the long-term mean monthly values are shown.

The correlations between long-term δ^{18} O values and the local surface air temperature are of great interest because of their possible application to paleoclimatic studies. The first relations of this kind were obtained by Dansgaard (1964) for mid- and high northern latitude coastal stations, and the slope was 0.69‰ per °C for δ^{18} O. Later, Rożanski *et al.* (1993) used the long-term monthly mean δ^{18} O_m and t_m values from the mid-latitude stations in the Northern Hemisphere to determine the slope of the best fit line of 0.31‰ per °C. This value is much lower than the one previously obtained by Dansgaard (1964), and it is in very good agreement with the slopes we obtained for stations Zagreb and Ljubljana (Eqs. 2–5). We should mention that the relationship δ^{18} O vs. t is linear only within a limited temperature range. Thus, the long-term seasonal correlation between δ^{18} O and surface air temperature derived for the stations of the IAEA/WMO global network situated between 40°N and 60°N showed that the δ^{18} O –t relationship is nonlinear, with a higher slope for lower δ^{18} O and t and only small changes of δ^{18} O with t temperatures (Rożanski 1993).

The mean yearly δ^{18} O in Zagreb is -9.1‰, and in Ljubljana -9.3‰. The peak-to-peak difference in δ^{18} O_m for both stations is ~6‰. Such a value is characteristic for continental stations (Vienna, Kraków) ~2000 km away from the main source of water, the Atlantic Ocean (Rożanski *et al.* 1993). Lower δ^{18} O values, especially during winter months, and larger amplitudes of seasonal δ^{18} O and δ^{2} H variations with increased distance from the ocean are usually described as the continental effect.

The large altitude difference between the station Puntijarka (988 m asl) and the Zagreb stations Grič and RBI (see Table 1) enables us to estimate the effect of altitude in our region. The difference in the mean yearly δ^{18} O values at Puntijarka and Zagreb-Grič is -2‰, and between Puntijarka and Zagreb-RBI is -2.76‰. Therefore, an average decrease in δ^{18} O value of -0.28‰ per 100 m altitude increase is estimated. The value is in good agreement with the global average value of -0.2% per 100 m altitude difference (Rozanski *et al.* 1993).

The close proportionality between $\delta^2 H$ and $\delta^{18} O$ in meteoric waters is commonly known as the global meteoric water line (GMWL):

$$\delta^2 H = 8 \,\delta^{18} O + 10\%. \tag{6}$$

This relation represents the base of many hydrological applications, because distinct deviations from the slope 8 indicate changes in isotopic composition due to evaporation, and the deviations from the GMWL indicate various isotopic exchange processes within an aquifer. The knowledge of the local meteoric water line (LMWL) is therefore of the utmost importance for any hydrogeological application of stable isotope data. We obtained the LMWLs for the station Zagreb-RBI (Fig. 10) by using all monthly data and by using the long-term monthly mean values:

all data:
$$\delta^2 H = (7.9 \pm 0.1) \, \delta^{18} O + (6.9 \pm 0.9)\%, \ n = 194, r = 0.985$$
 (7)

long-term:
$$\delta^2 H_m = (8.1 \pm 0.2) \, \delta^{18} O_m + (9.1 \pm 1.9)\%, \ n = 12, \ r = 0.996$$
 (8)

The corresponding LMWLs for Ljubljana are

all data:
$$\delta^2 H = (8.1 \pm 0.1) \, \delta^{18} O + (11.1 \pm 0.8)\%, \ n = 149, \ r = 0.992$$
 (9)

long-term:
$$\delta^2 H_m = (8.3 \pm 0.3) \, \delta^{18} O_m + (12.4 \pm 2.2)\%, \ n = 12, \ r = 0.995.$$
 (10)

The slope of the LMWL for Ljubljana is slightly higher than that for the Zagreb LMWL, and the intercept is somewhat higher. But the relations are not very different from those of the GMWL,

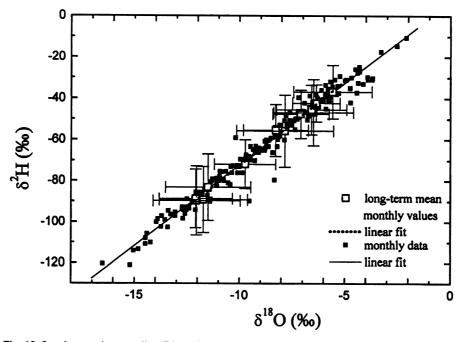


Fig. 10. Local meteoric water line (LMWL) for the Zagreb-RBI station. Black squares represent the monthly data, and white squares with error bars are the long-term mean monthly values ($\delta^{18}O_m$ and $\delta^{2}H_m$). Linear fits to all data and to the long-term mean monthly values are shown.

showing that the main source of precipitation is the Atlantic Ocean, and that no significant evaporation and isotope-exchange processes occur below the cloud (Rożanski et al. 1993).

The deviation of the individual data point from the GMWL is usually described as the deuterium excess d, where

$$d = \delta^2 H - 8 \times \delta^{18} O. \tag{11}$$

The deuterium excess reflects the specific source conditions during water vapor formation and it may therefore indicate different sources of vapor. For example, the precipitation originating from the Mediterranean Sea is characterized by distinctly higher deuterium excess than that originating from the Atlantic Ocean. The mean deuterium excess value for the Zagreb-RBI station is 7.8 ± 4.0 (n = 194), with almost normal distribution between -5 and +15. However, several monthly values are characterized by the values >20, indicating that the precipitation during these particular months originated predominantly from the Mediterranean Sea. No seasonal fluctuations in the deuterium excess are observed. For the Ljubljana station, the mean deuterium excess value is 9.7 ± 2.9 , distributed between 0 and 16, with no obvious seasonal variations, and no unusually high values.

CONCLUSION

In this paper, we presented the complete 20-yr-long record of environmental isotope data (²H, ³H, ¹⁴C, ¹⁸O) measured in Zagreb and collected at various stations in Croatia and Slovenia. The clean-air sites, the sites around the NPP and the sites in the industrialized area are included in the monitoring. Both radioactive isotopes (³H, ¹⁴C) show a continuous decrease in mean yearly values and superposed seasonal fluctuations. Relatively large peak-to-peak fluctuations in Δ^{14} C in the Zagreb area are caused by the introduction of fossil-fuel carbon into the atmosphere during the winter months.

A reconstruction of atmospheric ¹⁴C activity was done by measuring tree rings. A good correlation with the measured mean yearly Δ^{14} C in atmospheric CO₂ was obtained. The long-term ¹⁴C data (1964–1996) were fitted by the exponential decrease function, and the recent years (1983–1996) were fitted by a linear function. A linear function fitted to the mean yearly atmospheric ¹⁴C activity at Zagreb gave a decrease of 12.2‰ per year, and thus predicted that in 2000 in industrialized areas, the "natural" (pre-bomb–pre-industrial) level of Δ^{14} C = 0‰ can be reached, if no new sources of either very active or fossil-fuel ¹⁴C are introduced into the atmosphere.

In order to obtain approximate ³H values for hydrological applications, we reconstructed the past ³H activity in local precipitation by correlating our data with those from Vienna. The trend in mean, maximal and minimal yearly ³H activities in precipitation (1976–1996) is also analyzed by exponential decay functions. If no new ³H is introduced into the atmosphere in the future, then future ³H activity measured in precipitation should not be significantly different from the values measured in recent years. The predicted future ¹⁴C and ³H levels may later help determine any accidental contamination (local or global), or increased levels of continuous releases from various nuclear facilities. We found that atmospheric water vapor was very sensitive to local sources of ³H, and thus it can serve as a good indicator of local contamination. Simultaneous measurements of ³H concentration in precipitation and in atmospheric water vapor helped to identify the local contamination with ³H, and the new non-contaminated sampling sites for precipitation were found.

Long-term stable isotope records at two stations, Zagreb and Ljubljana, enabled us to determine local meteoric water lines, as well as the relation between δ^{18} O in precipitation and the local surface air temperature. Both relations are important for hydrological and paleoclimatic applications.

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