

ANTHROPOGENIC RADIOCARBON IN THE EASTERN IRISH SEA AND SCOTTISH COASTAL WATERS

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ABSTRACT. ¹⁴C is produced as an activation product in nuclear reactors, and may be discharged to both the atmosphere and the marine environment during nuclear fuel reprocessing. In the UK, ¹⁴C is discharged, under license, into the Eastern Irish Sea by the British Nuclear Fuels plc (BNFL) reprocessing plant at Sellafield, Cumbria, northwest England, and is then transported into Scottish coastal waters. We analyzed intertidal biota samples to determine the effect of these discharges. The specific activities of ¹⁴C found in these samples indicate that the uptake and bioaccumulation of ¹⁴C is dependent on the type of organism and its feeding behavior. Measured ¹⁴C concentrations in mussels (*Mytilus edulis*) were higher than those in winkles (*Littorina littorea*), which were greater than those found in seaweed (*Fucus* spp.); maximum observed activities were ca. 7, 5 and 3.5 times the accepted current ambient level of 260 Bq kg⁻¹ C, respectively. Annual Nori (*Porphyra umbilicalis*) samples were analyzed for their ¹³⁷Cs, ²⁴¹Am and ¹⁴C contents; both the ¹³⁷Cs and ²⁴¹Am results correlated well with published Sellafield discharge data ($r = 0.877$ and 0.918 , respectively), whereas there was no significant correlation between measured ¹⁴C activities and the discharge record, indicating increased complexity in the chemical and biological behavior of ¹⁴C or some discrepancy in the estimated discharge records.

INTRODUCTION

The natural equilibrium ¹⁴C concentration in the atmosphere, and subsequently, the oceans, has been perturbed over the last century by large injections of both stable and radioactive isotopes of carbon. A decrease of 2–3% due to fossil-fuel utilization by 1950 (Suess 1953, 1955) was overshadowed by inputs from nuclear weapons testing in the late 1950s and early 1960s ($2.2\text{--}3.5 \times 10^{17}$ Bq) (Lassey, Manning & O'Brien 1988; Taylor *et al.* 1990), from which time ~80% of the excess atmospheric ¹⁴C has been transferred to the surface and, ultimately, the deep oceans. Future inputs of anthropogenic ¹⁴C will depend on the growth and waste-management policies of the nuclear power industry. Current potential production is thought to be $\sim 4 \times 10^{14}$ Bq of ¹⁴C per year (Baxter, personal communication 1990; IAEA 1989), with 40% of this released directly into the atmosphere or the marine environment, and the remainder retained in the reactor until decommissioning. Many studies have been made on the likely discharges from nuclear establishments to the atmosphere (Kunz, Mahoney & Miller 1974, 1975; Magno, Nelson & Ellet 1975; Fowler *et al.* 1976; Moghissi & Carter 1977; Davis 1979; Beninson 1984; Kunz 1985) and the associated radiological implications to both local and global populations (Killough & Rohwer 1978; McCartney *et al.* 1986; McCartney 1987; McCartney, Baxter & Scott 1988a, b).

Although only a small proportion of the total ¹⁴C is discharged to the marine environment, ¹⁴C can be important for studying both the spatial distribution of a pollutant from a point source and the chemical behavior and fate of carbon within a near-shore environment. Ultimately, we can see the return of anthropogenic ¹⁴C to mankind *via* the marine food chain. Also, sources of anthropogenic ¹⁴C, such as Sellafield, in northwest England, offer a unique opportunity to study the carbon cycle in great detail on a local scale, with the potential for determining relative transfers and fluxes between the numerous global carbon reservoirs.

Although ¹⁴C is produced in small quantities relative to other radionuclides, such as the fission products, ⁹⁰Sr and ¹³⁷Cs, McCartney, Baxter and Scott (1988a) indicated that anthropogenic ¹⁴C will

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be radiologically significant on a global scale, delivering one of the highest collective effective dose equivalent commitments (CEDEC) to the global population, due to its long half-life and high mobility in the environment. Although discharge to the oceans reduces the dose slightly because of a longer residence time and, hence, more decay prior to human uptake (McCartney, Baxter & Scott, 1988a), ^{14}C will remain a radiologically important nuclide produced by the nuclear industry, unless alternative waste-management policies are implemented.

During nuclear fuel reprocessing at Sellafield, ^{14}C is released both to the atmosphere as $^{14}\text{CO}_2$ (~90%) and into the Eastern Irish Sea (~10%) via low-level liquid effluent discharges. The chemical form of the marine discharge is unknown but is thought to be a carbonate. The Sellafield site has been in operation since 1952, but has routinely monitored the amount of ^{14}C discharged only since 1 January 1984 (British Nuclear Fuels plc (BNFL) 1985). Discharge data published prior to this date are best-estimate values.

In an attempt to determine the fate of Sellafield-derived ^{14}C , we undertook a comprehensive study of the marine environment. We collected and analyzed a wide variety of samples, representing the biological and chemical forms present, to understand the processes occurring between them. Other areas of this study have focused on the distribution of ^{14}C in the water column, including the specific activities associated with the biogeochemical fractions (Begg *et al.* 1991). We are also studying phytoplankton, bottom-dwelling fish and intertidal and bottom sediments to determine the fate of Sellafield-derived ^{14}C . We focus here on the intertidal biota samples, and aim to answer three main questions:

1. Whether the study of intertidal biota and their utilization of different carbon pools as a source of nutrients can help determine the form of the ^{14}C discharged from Sellafield. Although our samples could also be used to calculate the possible radiological hazard to local populations consuming shellfish from the vicinity of Sellafield, our major goal is to understand the chemistry and cycling of this anthropogenic ^{14}C .
2. To measure the geographical extent of the transport of Sellafield-derived ^{14}C from the discharge location, using the selected organisms as indicators of the levels of ^{14}C either dissolved in the water column or as particulate material.
3. Whether previously estimated discharge data could be validated from a series of Nori (*Porphyra umbilicalis*) samples collected between 1967 and 1988. This would provide a more accurate record of past inputs into UK coastal waters and perhaps could be used in future mathematical models.

MATERIALS AND METHODS

Organism Selectivity as a Guide to the Chemical Form of Sellafield-Derived ^{14}C

We collected three organism types, mussels (*Mytilus edulis*), winkles (*Littorina littorea*) and seaweed (*Fucus* spp.), from 19 sites on a coastal transect from Earnse Point, 40 km south of Sellafield, north to Inverkip in the Clyde Sea area (Fig. 1), from November 1988 to February 1989. Mussels are filter feeders that retain particulate matter from the water column; winkles graze particulate matter deposited by seawater on large seaweeds and algae-covered rocks, whereas during photosynthesis, seaweed can directly utilize the inorganic carbon that is dissolved in the seawater. It is also feasible for seaweed to use atmospheric carbon during periods of low water.

Intra-site differences in ^{14}C concentrations between these species will reflect the specific activity of the carbon reservoirs that are the main sources of carbon for biological uptake; mussels will reflect the level of ^{14}C associated with the particulate matter suspended in the water column, where-

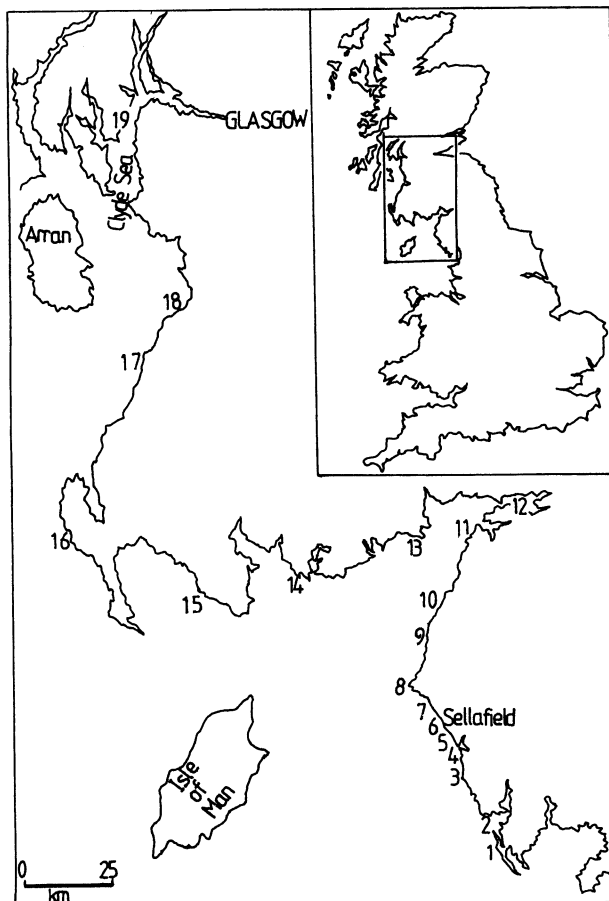


Fig. 1. Map of the study area showing the locations selected for intertidal biota sample collection November 1988–February 1989

as winkles will reflect the concentration of ^{14}C in the particulate matter deposited on surfaces by the surrounding water together with some seaweed-derived activity. Specific activities in seaweed should reflect those in the dissolved inorganic carbon (DIC) phase of seawater.

To complete this study, samples of shell material from both mussels and winkles from selected sites were also analyzed for their ^{14}C activity. Tanaka, Monaghan and Rye (1986) showed that 23–68% of molluscan shell material derived from metabolic carbon rather than DIC in the water column. Hence, Sellafield-derived ^{14}C should be incorporated into the shells of mollusks growing in the vicinity, and their specific activity will reflect the ^{14}C concentrations of both the surrounding DIC and that accumulated by the organism.

Spatial Distribution of Sellafield-Derived ^{14}C

Biota samples for the determination of organism selectivity were collected over an extensive geographical area to assess the spatial distribution of Sellafield-derived ^{14}C . We felt that this, in turn, would help us determine the chemical form of the ^{14}C during transport and the general behavior of carbon in the coastal marine environment.

Temporal Distribution of Sellafield-Derived ^{14}C

BNFL started monitoring and quantifying their discharges of ^{14}C from Sellafield in 1984. In an attempt to evaluate the release of ^{14}C to the Eastern Irish Sea prior to this date, the Ministry of

Agriculture, Fisheries and Food, Lowestoft (MAFF) submitted for analysis a suite of Nori samples collected at Seascale (2 km south of Sellafield) from 1967–1988, as part of their aquatic monitoring program. These samples were also analyzed for ^{137}Cs and ^{241}Am to ascertain whether they reflected the discharges associated with Sellafield. ^{137}Cs and ^{241}Am were selected, as both are anthropogenic (Sellafield being the only major source in this area), and exhibit two extreme forms of behavior in the water column – ^{137}Cs is known to be conservative and remain in the water column (Pierson 1988), whereas ^{241}Am is very particle-reactive and accumulates in areas of active sedimentation (Mackenzie, Scott & Williams 1987; McDonald *et al.* 1990).

Sample Analysis Procedures

In the laboratory, we separated mussel and winkle flesh from the shells. The flesh and seaweed samples were then washed in 0.5 M HCl to remove any extraneous carbonate material from their surfaces, rinsed in distilled water and dried to a constant weight at 60°C. For ^{14}C analysis, a representative 15–20 g of each sample were combusted in an oxygen-rich atmosphere in a high-pressure combustion vessel, as described by Barker, Burleigh and Meeks (1969). The resultant CO_2 was collected and converted to benzene, for liquid scintillation counting (LSC), *via* lithium carbide and acetylene (Barker 1953). The shell underwent an initial controlled hydrolysis to remove the outer 20% prior to ^{14}C analysis, to remove any extraneous material. The remaining material was hydrolyzed using 4 M HCl and, again, the resulting CO_2 was converted to benzene for LSC. When received from MAFF, the Nori samples were already dried and ground. Prior to combustion for ^{14}C analysis, 25 g of each sample were counted on a high-resolution reverse electrode germanium (HpGe) detector for the γ -ray emitting radionuclides, ^{137}Cs and ^{241}Am .

RESULTS AND DISCUSSION

All ^{14}C concentrations are reported as specific activities ($\text{Bq kg}^{-1} \text{ C}$), corrected for fractionation, relative to the international standard, NBS Oxalic Acid II. $\delta^{13}\text{C}$ values are determined relative to the international PDB standard used in all carbon isotope studies and are quoted as per mil (‰) differences from the standard. The biota sample results are also shown as $\Delta^{14}\text{C}$ values in the figures. The current accepted ambient ^{14}C concentration in the atmosphere is 260 $\text{Bq kg}^{-1} \text{ C}$, and although surface oceans are widely considered to have only 96% (NCRP 1985) of the atmospheric specific activity (due to differences in the rates of uptake and loss of $^{12}\text{CO}_2$, $^{13}\text{CO}_2$ and $^{14}\text{CO}_2$ across an air/water boundary), biota samples collected on the northwest coast of Scotland, where no anthropogenic inputs of ^{14}C occur, suggest that a baseline value of 260 $\text{Bq kg}^{-1} \text{ C}$ is realistic for this study of the environmental impact of Sellafield-derived ^{14}C on the marine system.

Organism Selectivity as a Guide to the Chemical Form of Sellafield-Derived ^{14}C

All the biota samples analyzed to date have shown ^{14}C concentrations above the baseline value of 260 $\text{Bq kg}^{-1} \text{ C}$ (Table 1). Figure 2 illustrates the ^{14}C concentrations found in the three sample types collected at the sites closest to Sellafield (locations 1–13). It is immediately apparent that the level of ^{14}C present in the organic material is greatly influenced by the type of organism analyzed.

At each site, mussels consistently showed the highest observed specific activities and seaweeds the lowest. This trend diminishes somewhat with distance from the point of discharge, as all three sample types appear to reach a common activity of 275–280 $\text{Bq kg}^{-1} \text{ C}$ at the sampling sites farther north within the Clyde Sea area. This common activity is not observed at the southerly sampling stations, where organism selectivity is still apparent. Extrapolation of Figure 2 would indicate that a common specific activity would be obtained if the sampling transect was extended to 50–60 km from Sellafield.

TABLE 1. Specific Activities ($\text{Bq kg}^{-1} \text{ C} \pm 1\sigma$) Measured in Coastal Biota Samples

| Site no. | Location | Distance | Specific Activity ($\text{Bq kg}^{-1} \text{ C} \pm 1\sigma$) | | |
|----------|-------------------|---------------|---|--------------------|-------------------|
| | | | Mussels | Winkles | Seaweed |
| 1 | Earnse Point | 37.8 km South | 601.15 ± 2.00 | 501.28 ± 1.88 | 352.95 ± 1.54 |
| 2 | Haverigg | 30.3 km South | NA | NA | 356.66 ± 1.61 |
| 3 | Annaside | 19.8 km South | NA | NA | 433.28 ± 1.87 |
| 4 | Ravenglass | 10.8 km South | 1542.91 ± 5.94 | NA | 756.41 ± 3.43 |
| 5 | Drigg | 6.5 km South | 1708.73 ± 4.93 | 1386.22 ± 4.44 | NA |
| 6 | Sellafield | 0 | 1779.10 ± 6.58 | 983.68 ± 4.23 | 937.42 ± 3.38 |
| 7 | Nethertown | 5 km North | 1412.62 ± 4.20 | 988.42 ± 4.66 | 527.35 ± 3.41 |
| 8 | St. Bees | 10.2 km North | NA | NA | 469.61 ± 1.45 |
| 9 | Workington | 30 km North | NA | 453.32 ± 2.17 | 343.33 ± 2.20 |
| 10 | Maryport | 38.5 km North | NA | NA | 331.20 ± 2.02 |
| 11 | Silloth | 52 km North | NA | NA | 325.47 ± 1.18 |
| 12 | Bowness-on-Solway | 58 km North | NA | NA | 286.55 ± 3.38 |
| 13 | Rockcliffe | 77 km North | 449.04 ± 1.89 | 330.80 ± 1.28 | 297.81 ± 1.08 |
| 14 | Auchenlarie | | 401.07 ± 1.57 | 348.24 ± 1.30 | 311.23 ± 1.71 |
| 15 | Port William | | NA | NA | 314.91 ± 1.25 |
| 16 | Port Patrick | | NA | NA | 281.91 ± 1.06 |
| 17 | Girvan | | 281.96 ± 1.30 | NA | 283.75 ± 1.06 |
| 18 | Ayr | | 272.48 ± 0.94 | 280.69 ± 1.25 | 274.87 ± 0.96 |
| 19 | Inverkip | | 281.96 ± 1.30 | 280.18 ± 1.10 | 272.99 ± 1.16 |

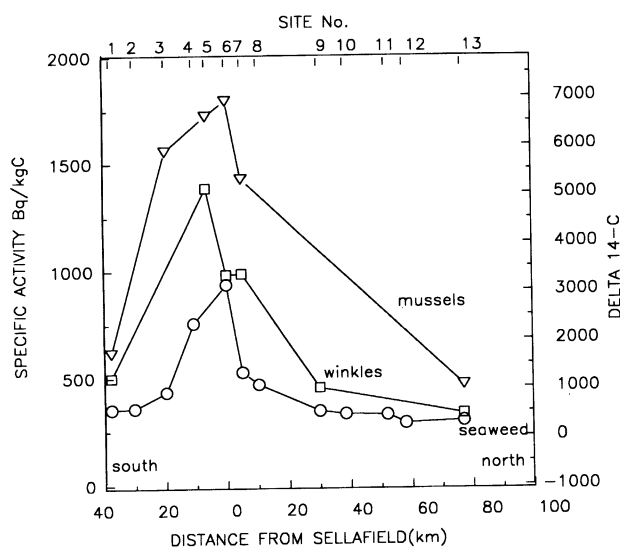


Fig. 2. Specific activities measured in biota samples collected along a coastal transect in the immediate vicinity of Sellafield

The highest observed specific activities for mussels ($1779 \text{ Bq kg}^{-1} \text{ C}$) and seaweed ($937 \text{ Bq kg}^{-1} \text{ C}$) were found at Sellafield, whereas that for winkles ($1386 \text{ Bq kg}^{-1} \text{ C}$) was found 6.5 km to the south at Drigg. These values correspond to specific activities *ca.* 7, 3.5 and 5 times the accepted current baseline value of $260 \text{ Bq kg}^{-1} \text{ C}$, and indicate some degree of organism selectivity occurring in the uptake of anthropogenic ^{14}C due to differences in feeding behavior. ^{14}C enhancements were also found in both mussel and winkle shells, but these were less than those found in the

corresponding organic matter, *e.g.*, 1207 Bq kg⁻¹ C_{inorg} and 1779 Bq kg⁻¹ C_{org} in Sellafield mussels, 1100 Bq kg⁻¹ C_{inorg} and 1386 Bq kg⁻¹ C_{org} in Drigg winkles. The lower activity in the inorganic matter may reflect the fact that the shell carbon accumulates throughout the entire lifespan of the organism, whereas the organic tissue is subject to turnover at a rate that is organism-specific. Tanaka, Monaghan and Rye (1986) found that the mollusk shells reflected the incorporation of metabolic organic carbon in conjunction with uptake of DIC from the water column. Hence, this may also explain the differences found between the inorganic and organic carbon specific activities in our samples. From these results, it appears that the discharges of ¹⁴C from Sellafield are mainly, if not solely, in a dissolved inorganic form, which can be taken up by biota in the vicinity and incorporated into both the inorganic shell and the organic flesh of the organism. Further analyses on biogeochemical phases of the water column support this interpretation, as enhancements of ¹⁴C are not found in either the dissolved or particulate organic carbon fractions, but levels above the current ambient background are observed in the DIC fraction (Begg *et al.* 1991).

From 1952 to 1984, when actual discharges were monitored, ¹⁴C discharges from Sellafield have appeared to be relatively constant (R. Atherton, personal communication 1988). Thus, differences in the lifespans of the three biota types are unlikely to be major contributory factors to the observed activity trends. Hence, the results indicate that the specific activity of ¹⁴C in the particulate phase of the water column must be higher than that in the DIC. We have found no evidence of this during the study of the biogeochemical phases in the water column (Begg *et al.* 1991). However, the water samples were collected in December 1989 during a winter sampling cruise, and the source of ¹⁴C-enriched particulate matter may be seasonally dependent, *i.e.*, phytoplankton that photosynthesize in the general Sellafield area during the summer months. Although the observed specific activities in the DIC (268–575 Bq kg⁻¹ C) do not approach those found at corresponding sites in the mollusks, one must remember that the water was sampled and analyzed at one time only, whereas the biota represent an integrated accumulation throughout their lifetimes.

Spatial Distribution of Sellafield-Derived ¹⁴C

The effect of ¹⁴C discharges from Sellafield can be seen as far north as the Clyde Sea area, although the interorganism variation is negligible. Figure 2 shows the results obtained for the three organisms collected from 40 km south to 80 km north, indicating that, although the highest levels are found in the immediate vicinity of Sellafield, movement of the radionuclide does occur both to the north and south of the release point. For mussels and seaweed, the specific activities sharply decrease directly to the north of Sellafield – much more obviously than to the south – indicating that the transport of ¹⁴C in the water column reflects the predominant residual surface currents in the area. These tend to follow the coastline south before turning to flow northward out of the Eastern Irish Sea *via* the North Channel (Dickson & Boelens 1988). In the case of winkles, where the maximum activity is found south of Sellafield, this mirroring of the local currents is more pronounced, although more sampling locations need to be considered to determine the rate of decline in activity to the south. This similarity in distribution points to a common source of ¹⁴C for all three organisms within the study area.

We considered two parameters in an attempt to model empirically this observed variation, namely, the distance and direction from the point of release. For the mussels and winkles, >90% of the observed variation could be explained in terms of distance alone. For seaweed, the concentrations were best explained when both distance and direction were taken into account, which may indicate the use of atmospheric CO₂ during periods of low water.

TABLE 2. Sellafield Discharges and Measured Activities in Nori (*Porphyra*) at Seascale

| Year | ^{14}C | | | ^{137}Cs | | | ^{241}Am | | |
|------|---|---------------------|---|---|---------------------|---|---------------------|---|---------------------|
| | Activity (Bq kg $^{-1}$ C \pm 1 σ) | Discharges (TBq) | Activity (mBq g $^{-1}$ \pm 1 σ) | Activity (mBq g $^{-1}$ \pm 1 σ) | Discharges (TBq) | Activity (mBq g $^{-1}$ \pm 1 σ) | Discharges (TBq) | Activity (mBq g $^{-1}$ \pm 1 σ) | Discharges (TBq) |
| 1967 | 379.07 \pm 1.53 | 0.50 | 34.51 \pm 2.42 | 40.61 \pm 9.54 | 150 | 40.61 \pm 9.54 | 17.09 | | |
| 1968 | 451.52 \pm 1.55 | 0.75 | 131.50 \pm 3.02 | 528.09 \pm 3.17 | 371 | 528.09 \pm 3.17 | 21.13 | | |
| 1969 | 541.35 \pm 3.48 | 0.83 | 175.86 \pm 3.34 | 450.13 \pm 5.40 | 444 | 450.13 \pm 5.40 | 14.43 | | |
| 1970 | 556.48 \pm 1.82 | 1.1 | 609.20 \pm 6.09 | 68.66 \pm 2.65 | 1154 | 68.66 \pm 2.65 | 19.09 | | |
| 1971 | 487.31 \pm 1.71 | 1.2 | 753.85 \pm 6.78 | 407.79 \pm 5.30 | 1325 | 407.79 \pm 5.30 | 37.67 | | |
| 1972 | 387.03 \pm 1.57 | 1.4 | 405.84 \pm 4.46 | 1290.69 \pm 5.16 | 1289 | 1290.69 \pm 5.16 | 79.51 | | |
| 1973 | 712.86 \pm 2.02 | 0.9 | 874.25 \pm 6.12 | 1269.73 \pm 12.70 | 768 | 1269.73 \pm 12.70 | 109.19 | | |
| 1974 | 372.75 \pm 1.41 | 0.79 | 2355.85 \pm 9.42 | 1661.14 \pm 4.98 | 4061 | 1661.14 \pm 4.98 | 118.25 | | |
| 1975 | 461.84 \pm 1.69 | 1.1 | 1310.58 \pm 7.86 | 698.45 \pm 23.05 | 5231 | 698.45 \pm 23.05 | 36.25 | | |
| 1976 | 333.10 \pm 1.44 | 1.1 | 1586.87 \pm 7.93 | 202.87 \pm 3.65 | 4289 | 202.87 \pm 3.65 | 11.95 | | |
| 1977 | 421.34 \pm 1.39 | 0.77 | 2759.18 \pm 11.04 | 502.52 \pm 18.56 | 4478 | 502.52 \pm 18.56 | 3.66 | | |
| 1978 | 449.36 \pm 1.73 | 0.83 | 1715.32 \pm 8.58 | 297.08 \pm 4.75 | 4088 | 297.08 \pm 4.75 | 7.92 | | |
| 1979 | 471.60 \pm 1.48 | 1.1 | 1192.83 \pm 8.35 | 359.94 \pm 23.04 | 2562 | 359.94 \pm 23.04 | 7.84 | | |
| 1980 | 1010.18 \pm 2.77 | 1.1 | 1294.79 \pm 9.06 | 289.34 \pm 4.63 | 2966 | 289.34 \pm 4.63 | 8.25 | | |
| 1981 | 678.61 \pm 2.18 | NA | 650.24 \pm 5.85 | 559.64 \pm 52.61 | 2357 | 559.64 \pm 52.61 | 8.77 | | |
| 1982 | 537.60 \pm 1.62 | NA | 277.09 \pm 4.99 | 77.28 \pm 2.40 | 2000 | 77.28 \pm 2.40 | 6.4 | | |
| 1983 | 602.13 \pm 2.09 | NA | 357.83 \pm 4.29 | 98.30 \pm 19.56 | 1200 | 98.30 \pm 19.56 | 2.22 | | |
| 1984 | NA | 0.7 | 280.36 \pm 3.64 | 98.90 \pm 13.06 | 434 | 98.90 \pm 13.06 | 2.3 | | |
| 1985 | 544.32 \pm 1.82 | 1.3 | 233.94 \pm 3.81 | 96.23 \pm 19.34 | 325 | 96.23 \pm 19.34 | 1.6 | | |
| 1986 | 527.46 \pm 1.93 | 2.6 | 204.28 \pm 2.86 | 53.15 \pm 2.07 | 17.9 | 53.15 \pm 2.07 | 1.3 | | |
| 1987 | 462.36 \pm 1.59 | 2.1 | 68.89 \pm 1.86 | 58.08 \pm 2.61 | 11.88 | 58.08 \pm 2.61 | 0.65 | | |
| 1988 | 606.63 \pm 1.91 | 3.0 | 42.73 \pm 1.58 | 62.29 \pm 2.12 | 13.3 | 62.29 \pm 2.12 | 0.75 | | |

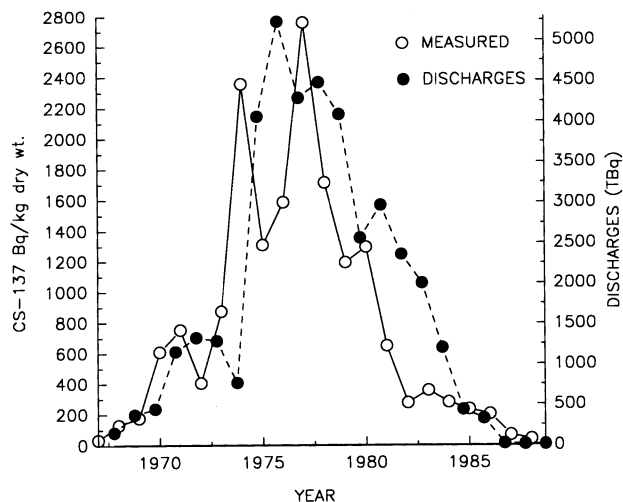


Fig. 3A. ^{137}Cs activities in the annual *Porphyra* samples collected at Seascale 1967–1988 shown with the published ^{137}Cs discharge data from Sellafield

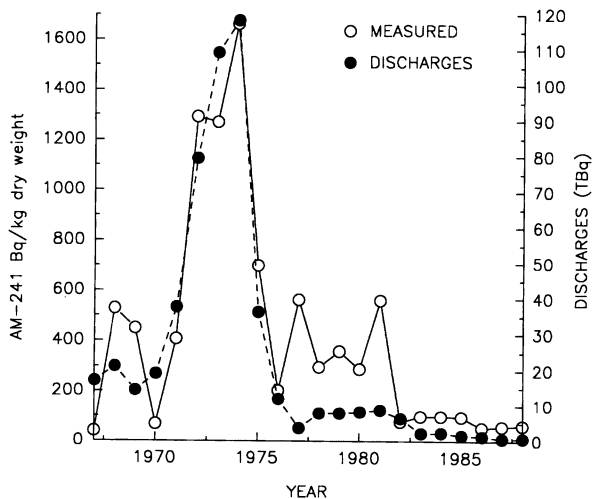


Fig. 3B. ^{241}Am activities measured in the annual *Porphyra* samples collected at Seascale 1967–1988 shown with the published ^{241}Am discharge data from Sellafield

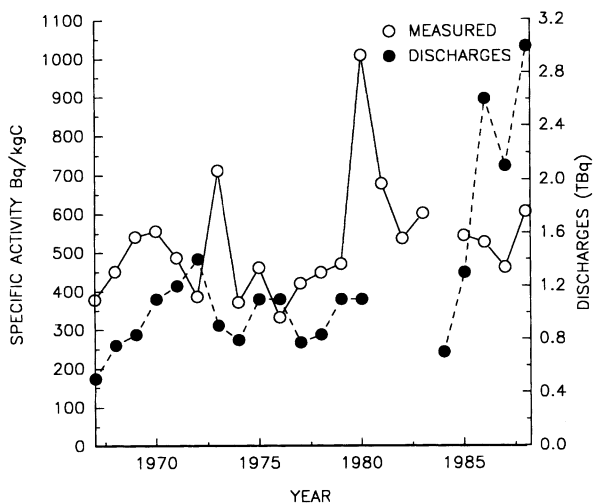


Fig. 3C. ^{14}C specific activities measured in the annual *Porphyra* samples collected at Seascale 1967–1988 shown with the published ^{14}C discharge data from Sellafield

Temporal Distribution of Sellafield-Derived ^{14}C

Sellafield discharges a wide variety of both activation and fission products that have no natural production pathway and exhibit a varied pattern of chemical behavior. We measured two such anthropogenic radionuclides, ^{137}Cs (conservative) and ^{241}Am (particle-reactive), to determine the suitability of the chosen samples for a temporal study. Both have shown excellent agreement between the published discharge data and the activities measured using γ -ray spectroscopy (Table 2; Figs. 3A, B). The correlation coefficients (r) obtained for the measured activities and their relevant discharge data were 0.877 and 0.918, respectively, which indicate that the Nori samples significantly reflect the levels of discharge in any given year for these particular nuclides.

Figure 3C illustrates the relationship observed between the ^{14}C specific activities as measured in the Nori samples and the discharge data released by BNFL. The correlation coefficient obtained in this case ($r = 0.153$) is not significant, which precludes these samples from being used to reconstruct the discharge record of ^{14}C into the marine environment from Sellafield.

Inclusion of influences from the atmosphere and hence, the bomb-produced ^{14}C , over this time period does not improve the correlation coefficient, but problems still arise in matching the observed activities to the published discharge data. Previous work has shown that the activities measured in seaweed are considerably greater than those measured in the DIC fraction of the water column (Begg *et al.* 1991), indicating a bio-accumulation effect, which is likely to overshadow small variations resulting from differences in the levels of discharge.

To minimize seasonality effects, all samples were collected from March through June in each year. However, there is no information on the regularity or timing of ^{14}C discharges. Because ^{14}C is obviously much more "biologically involved" than either ^{137}Cs or ^{241}Am , the seasonality of the discharges may be more important and, thus, could partially account for the poor correlation.

CONCLUSIONS

Biota samples from an area 40 km south of Sellafield northwards into the Clyde Sea are enriched in ^{14}C . A common activity 15–20 Bq kg $^{-1}$ C above the accepted baseline value is reached at sites farthest from Sellafield. Despite the fact that the activities in mussels (exclusively filter feeders) are the highest, we found no evidence of enriched particulate carbon in the water column (Begg *et al.* 1991). This suggests that the particulate matter is seasonal, *i.e.*, phytoplankton, and work is now in progress to obtain and analyze such material. Of course, we do not rule out the possibility that some of the Sellafield discharge is in the form of particulate carbon, although it appears that most of the discharged ^{14}C enters the DIC phase of the water column.

Although both ^{137}Cs and ^{241}Am measured in annually collected Nori showed excellent correlations with published Sellafield discharges, no such correlation was observed with ^{14}C , thus precluding the use of these samples to help reconstruct past ^{14}C discharges. This is likely to be a reflection of the more complex biological involvement of ^{14}C rather than discrepancies in the reported discharge data, although the latter cannot be totally ruled out.

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