ATMOSPHERIC RADIOCARBON WORKSHOP REPORT

Jocelyn Turnbull^{1,2} • Heather Graven³ • John Miller^{2,4} • Scott Lehman⁵ • Workshop Participants

ABSTRACT. A workshop on atmospheric radiocarbon measurements was held in conjunction with the 21st International Radiocarbon conference in July 2012. The main topics were intercomparison of measurements of ¹⁴C in atmospheric CO₂, the potential for use of gas standards for atmospheric ¹⁴C measurements, reporting of uncertainties, and expansion of intercomparison activities to other atmospheric trace gases. This report documents the discussions and conclusions of this workshop.

INTRODUCTION

Measurements of radiocarbon in atmospheric trace gases are becoming more widely used for a variety of scientific applications. There is, in particular, a growing focus on the use of the ¹⁴C content of carbon dioxide (14CO₂) as a tracer for fossil-fuel CO₂ emissions, which requires high-precision measurement and comparability, as the ¹⁴C measurement precision is often close to the size of the atmospheric signals of interest. Both the ¹⁴C and atmospheric trace gas measurement communities have long histories of intercomparison activities to ultimately allow measurements from different laboratories to be used as a single data set. The World Meteorological Organization (WMO) holds a biannual meeting on Greenhouse Gas Measurement Techniques (GGMT), and publishes biannual Expert Group Recommendations for measurement, reporting, and intercomparison of greenhouse gases and related tracers (WMO 2011). ¹⁴C in atmospheric trace gases has been included in these recommendations since 2005. Only a small subset of the atmospheric ¹⁴C community attended the 2011 GGMT meeting in Wellington, New Zealand, and a more detailed discussion was felt to be necessary. Therefore, a workshop on atmospheric ¹⁴C measurements was convened as part of the 21st International Radiocarbon Conference. This report documents the discussion and recommendations from the workshop, which will also be included in the next WMO/GGMT recommendations due to be published in 2013.

The major topics of discussion were ongoing and new intercomparison activities for ¹⁴CO₂ measurements, the use of gas standards for accelerator mass spectrometry (AMS) ¹⁴CO₂ measurements, reporting of uncertainties for atmospheric ¹⁴C measurements, and the need for intercomparison activities for ¹⁴C in other atmospheric trace gases.

INTERCOMPARISON ACTIVITIES

Ongoing Flask ¹⁴CO₂ Intercomparison

John Miller (NOAA/ESRL) reported on the results from all 3 rounds of the $^{14}\text{CO}_2$ flask intercomparison, which was started in 2007 (Miller et al. 2011, 2013). This intercomparison is presently limited to AMS laboratories, which are capable of making measurements from relatively small volumes of whole air (i.e. yielding ~40–100 µmole CO_2). Contributors send between 2 and 4 air sampling flasks to the University of Colorado to be filled with whole air from 2 high-pressure gas cylinders: FARI-A and FARI-B. FARI-A is ambient air filled at Niwot Ridge, Colorado. FARI-B is ambient air

¹GNS Science, Lower Hutt, New Zealand. Corresponding author. Email: j.turnbull@gns.cri.nz.

²CIRES, University of Colorado, Boulder, Colorado, USA.

³Scripps Institution of Oceanography, San Diego, California, USA.

⁴NOAA Earth System Research Laboratory, Boulder, Colorado, USA.

⁵INSTAAR, University of Colorado, Boulder, Colorado, USA.

that has been spiked with a small amount of 14 C-free CO₂ to obtain an air mixture with CO₂ mole fraction enhanced by ~40 µmole/mole and with a 14 C content (14 C/C ratio) slightly below ambient.

The results show reasonable agreement between the participating laboratories but more spread than expected, and are reported in full in Miller et al. (these proceedings). Some of the spread between results from different laboratories may be related to choice of material used for standardization (Ox I vs. Ox II). This issue needs further investigation. Measurements of CO_2 mole fraction and $\delta^{13}C$ were made by some of the laboratories, and it appears that variability in filling of the flasks cannot explain the variability in $\Delta^{14}CO_2$. However, measurement of CO_2 mole fraction and $\delta^{13}C$ may still help to understand the source of interlaboratory variability.

The workshop participants discussed the experimental design of the previous intercomparison and ideas for improvements for future rounds. The current flask filling method is to first "condition" the flasks by evacuating, then twice filling all flasks with the cylinder air and re-evacuating again, then filling flasks to slightly over ambient pressure with the cylinder air for distribution to the labs. This method was deemed to be adequate for the \$^{14}CO_2\$ intercomparison. The \$^{13}CO_2\$ intercomparison exercise uses a more rigorous flushing method whereby all flasks are connected in series and air from the cylinder is flushed through the "sausage" of flasks for a period of time before closing off all flasks for distribution to the labs (e.g. Mukai et al. 2005). This method may use too much air, and is unnecessary for the level of precision of the \$^{14}CO_2\$ intercomparison exercise. In particular, if flushing is done, a new tank is needed for every intercomparison, whereas the evacuation method currently in use allows the same tanks to be used for multiple intercomparison rounds. There remains sufficient air in these tanks for several more rounds at least.

Currently, additional control flasks are filled and measured at the University of Colorado and NOAA/ESRL for CO_2 mole fraction and $\delta^{13}CO_2$. It was agreed that all the future intercomparison flasks be measured for CO_2 mole fraction and $\delta^{13}CO_2$, and that the $\delta^{13}C$ measurement be of sufficient quality to detect variability of ~0.1% in $\delta^{13}C$. It has not yet been agreed whether individual participants will make these measurements, or if they can be performed at the University of Colorado and NOAA/ESRL. A second intercomparison exercise, to be based on a distribution of pure CO_2 produced from the same source, was proposed, so that differences arising from CO_2 extraction procedures and graphitization/AMS measurements can be isolated. Two options for generating this CO_2 were discussed: CO_2 extracted at the University of Colorado from the existing FARI tanks; and obtaining modern CO_2 from a source such as ethanol fermentation. Pure CO_2 would be distributed in flame-sealed ampoules.

AMS and Proportional Counting Intercomparison

A major challenge is to compare atmospheric $^{14}\text{CO}_2$ results for a common source of air for both gas counting/liquid scintillation and AMS labs, as gas counting and liquid scintillation require much larger amounts of sample material than can be obtained from the ~2- to 5-L flasks of air now used for AMS. Some intercomparison of this type has already been done on a limited basis. The University of Heidelberg has an ongoing in-house comparison whereby they collect CO_2 absorbed onto NaOH for gas counting and simultaneously collect air from the same inlet line into polyethylene-coated aluminum bags. Aliquots from the bags are then measured by AMS (Vogel et al. 2011).

Ingeborg Levin and Samuel Hammer from the Heidelberg group volunteered to provide CO₂ aliquots from the NaOH absorption for measurement in other AMS labs. Although this does not capture possible variability arising from interlaboratory differences in the extraction process, it will nevertheless provide for a comparison of the ¹⁴CO₂ measurement scale between labs.

Co-Located Sampling Intercomparisons

Heather Graven (Scripps Institution of Oceanography) and Xiaomei Xu (University of California, Irvine) presented a comparison of results from 2 different flask sampling programs at Point Barrow, Alaska (Graven et al., these proceedings). They showed very good agreement between the 2 independent sampling and measurement programs, both using AMS measurement (Graven et al., these proceedings).

Co-located sampling was discussed as a further intercomparison exercise, but the participants felt that although this is desirable, it would be challenging to do this across all laboratories. Instead, it is recommended that this continue to be done when feasible as bilateral arrangements between labs rather than as a community exercise.

Future Intercomparison Rounds

The next round (round 4) of atmospheric ¹⁴CO₂ intercomparison is currently being planned. Future intercomparison rounds are expected to be approximately once every 1–2 yr, with the timing dependent on receiving flasks from the contributing laboratories and reporting of measurement results.

Round 4 will include 3 different intercomparison activities as discussed above. Participants are encouraged, but not required, to participate in all 3 intercomparisons:

- A. Flasks of whole air for extraction, filled at University of Colorado. Only available to AMS labs. This is the same as the round 1–3 intercomparisons, except that CO₂ mole fraction and δ¹³CO₂ measurements will be included.
- B. Flame-sealed ampoules of CO₂ from Heidelberg (Levin/Hammer), for direct comparison with the Heidelberg gas counting system. This could also include larger samples, enabling other liquid scintillation and gas counting labs to participate.
- C. Flame-sealed ampoules of CO₂ from University of Colorado (Lehman/Miller) from the same tanks as the whole air samples. Only available to AMS labs due to the limited size of tanks.

Any interested contributors who are not already involved should contact John Miller (john.b.miller@noaa.gov) for further information regarding participation. This intercomparison work does not have any direct funding, and the cost of preparation of samples is currently being carried by the University of Colorado and NOAA. Funding support, even at very small levels, would improve these intercomparisons, and any suggestions for funding avenues should be directed to John Miller.

USE OF GAS STANDARD VS. SOLID STANDARDS FOR ATMOSPHERIC 14C MEASUREMENTS

The choice of standard material used in 14 C measurement was discussed. The preparation method for oxalic acid I and II is to combust and purify the solid material, which appears to introduce more variance than is seen in CO_2 extracted from whole air (Graven et al. 2007; Graven 2008). This variance may be associated with small batch combustion of the solid materials. Preparation of large batches of CO_2 from the solid materials may help with this, but there is also concern that a large batch may drift over time or have sudden shifts in 14 C value, and liquid CO_2 (at pressure >60 bar) cannot be allowed to form in the storage container.

CO₂ extracted from high-pressure cylinders of natural whole air has been used as a working reference for atmospheric measurements (Graven 2008). The mean value of CO₂ in the whole air tank is characterized by normalizing it to Ox I over multiple AMS measurement runs. The gas standard is

then used to normalize authentic samples and "surveillance" gas standards in each individual wheel (Graven 2008).

Concerns about natural air standards are that they could potentially drift through time, and have the logistical challenge that cylinders will be used up fairly quickly. Participants agreed that moving completely away from the solid standards would be a mistake, and that the atmospheric ¹⁴C community needs to remain tied to Ox I and/or Ox II, even if a reference gas material is also used as a working reference.

A small working group was formed to investigate the development of a gas standard for 14 C, aiming to collaborate with other entities (such as the US National Institute of Standards and Technology) to help develop such a standard. Scott Lehman and Tom Guilderson will organize and lead this group. One major task of this working group will be to resolve the question of whether such a gas standard should be provided as whole air or as CO_2 .

REPORTING OF UNCERTAINTIES IN ATMOSPHERIC 14C MEASUREMENTS

As atmospheric ¹⁴C measurements are at the forefront of ¹⁴C measurement precision, reporting of uncertainties has received considerable attention. At the sub-2‰ level of precision now often reported for ¹⁴CO₂ measurements, the common practice in the ¹⁴C community of reporting counting statistical uncertainties as the full uncertainty is unlikely to capture the entire measurement uncertainty (Miller et al., these proceedings). Therefore, it is recommended that reporting of atmospheric ¹⁴C results should include uncertainties that consider long-term repeatability of authentic atmospheric materials. Repeatability is typically determined as the scatter of long-term repeated measurements of authentic atmospheric materials such as extraction aliquots from surveillance cylinders of whole air or replicates of atmospheric samples.

INTERCOMPARISON ACTIVITIES FOR 14C IN OTHER ATMOSPHERIC TRACE GASES

There is some interest in developing intercomparisons for ¹⁴CH₄ and ¹⁴CO, as several laboratories are now making one or both of these measurements. A major difficulty is that large amounts of air are needed for measurement. The consensus was that bilateral intercomparisons should be encouraged, and that the need for a formal intercomparison will be re-evaluated at the next WMO GGMT meeting.

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