# Restoring riparian corridors with fire: Effects on soil and vegetation

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#### **Abstract**

In many riparian corridors of the semi-arid west, stream incision has resulted in lowered water tables, basin big sagebrush (Artemisia tridentata var. tridentata Nutt.) encroachment and the loss of the dominant herbaceous vegetation. To determine the potential for restoring basin big sagebrush-dominated riparian corridors to greater herbaceous cover, a fall prescribed burn on sites with relatively shallow (-153 to -267 cm) and deep (-268 to  $\geq$  -300 cm) water tables was conducted. We evaluated the separate and interacting effects of water table depth and burning on total soil C and N, soil nutrient availability, and soil enzyme activities by microsite (sagebrush subcanopy, sagebrush interspace), and soil depth (ash/liter, 0-2, 2-5, and 10-20 cm). Three years after the prescribed burn, tissue nutrient content in silvery lupine (Lupinus argenteus Pursh) and Douglas sedge (Carex douglasii Boott), by microsite, growing in burned and unburned areas of 1 shallow water table site was measured. Influence of fire on soil attributes was largely limited to the top 5 cm. As a consequence of prescribed burning, deep water table sites lost relatively more N and C from litter horizons and released more aqueous-soluble Ca+2 from 0-2 cm mineral horizons than did corresponding horizons from shallow water table sites. Overall, compared to unburned controls, burning: (1) increased aqueousextractable SO<sub>4</sub><sup>-2</sup>, K<sup>+</sup>, and KCl-extractable NH<sub>4</sub><sup>+</sup>, (2) decreased activities of the enzymes asparaginase, urease and acid-phosphatase, and (3) decreased KCl-extractable NO<sub>3</sub> and aqueoussoluble ortho-P. Out of 16 measured soil attributes reported, 7 were influenced by a microsite main effect and/or interaction. New tissue of silvery lupine on burned plots had greater N, greater Zn and Fe (only on subcanopy microsites) and less Ca than plants on control plots; new tissue of Douglas sedge had greater S and less Na, P, and Zn. The results indicate that burning alone is an appropriate restoration treatment for shallow water table sites because of minimal C and N loss and increased available nutrients for regrowth of understory herbaceous species. Deep water table sites require a burning prescription that minimizes fire severity because of higher potential C and N loss, and reseeding due to a lack of perennial understory herbaceous species and more xeric conditions.

Key Words: restoration, riparian ecosystems, water table depth

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### Resumen

En muchos corredores ribereños del oeste semiárido la incisión de corrientes ha resultado en mantos freáticos mas bajos, la expansión de "Basin big sagebrush" (Artemisia tridentata var. tridentata Nutt.) y la perdida de vegetación herbácea dominante. Para determinar el potencial para restaurar los corredores ribereños dominados por "Basin big sagebrush con una mayor cobertura herbácea se llevo a cabo un fuego prescrito en otoño en sitios con mantos freáticos relativamente poco profundos (-153 to -267 cm) y mantos freáticos profundos (-268 to  $\geq$  -300 cm). Evaluamos los efectos separados e interactuantes de la profundidad del manto fréatico y la quema en el C y N total del suelo, la disponibilidad de nutrientes del suelo y las actividades enzimáticas del suelo por micrositio ("Sagebrush" subcopa, "Sagebrush" interespacio) y profundidad del suelo (ceniza /mantillo, 0-2, 2-5 y 10-20 cm). Tres años después del fuego prescrito se midió por micrositio, el contenido de nutrientes de los tejidos de "Silvery lupine" (Lupinus argenteus Pursh) y "Douglas sedge" (Carex douglasii Boott), creciendo en áreas quemadas y sin quemar de un sitio con manto fréatico somero. La influencia del fuego en los atributos del suelo se limito principalmente a los 5 cm superiores. Como consecuencia del fuego prescrito, los sitios con manto fréatico profundo perdieron relativamente más N y C de los horizontes con mantillo y liberaron mas Ca<sup>+2</sup> soluble en agua de los horizontes minerales de 0-2 cm que los horizontes correspondientes en los sitios con manto fréatico poco profundo. En general, comparado con los controles sin quemar, la quema (1) incremento el SO<sub>4</sub>-2 extractables en agua, K+ y KClextractable NH<sub>4</sub><sup>+</sup>, (2) Disminuyo la actividad de las enzimas asparaginasa, ureasa y ácido- fosfatasa y (3) disminuyo el KClextractable, NO<sub>3</sub> y los ortho-P soluble en agua. De los 16 atributos de suelo medidos, 7 fueron influenciados por un efecto principal de micrositio y/o interacción. El tejido nuevo de "Silvery lupine"en las parcelas quemadas tuvo mayor contenido de N. Zn y Fe (solo en los micrositios de abajo de la copa) y menos Ca que las plantas en las parcelas control; el nuevo tejido del "Douglas sedge"tuvo más S y menos Na, P y Zn. Z. Los resultados indican que la quema sola es un tratamiento apropiado para restauración de sitios con agua fréatica somera debido a las perdidas mínimas de C y N y el aumento en la disponibilidad de nutrientes para el rebrote de especies herbáceas. Los sitios con manto fréatico profundo requieren de una prescripción de fuego que minimice la severidad del fuego por el alto potencial de perdida de C y N y la resiembra debido a la falta de especies herbáceas perennes y a condiciones mas xéricas.

Streams in the central Great Basin are in an incisional phase (Miller et. al. 2001), that is being exacerbated by anthropogenic disturbance (Chambers et al. 1998). Stream incision often results in lowered water tables and encroachment and dominance of basin big sagebrush (Artemisia tridentata var. tridentata Nutt.) in riparian corridors, and the loss of grass and forb-dominated meadow ecosystems. For riparian meadow ecosystems in general, the hydrologic regime, as indicated by water table depth, determines both soil properties (Chambers et al. 1999) and riparian ecosystem characteristics (Castelli et al. 2000). Thus, the potential for restoring sagebrush-dominated ecosystems to grass and sedge meadows depends on whether or not an abiotic threshold defined by water table depth has been crossed (Chambers and Linnerooth 2001). The characteristics of the soils and their response to restoration treatments are critical determinants of the restoration potential.

A restoration experiment was conducted to determine the potential for restoring basin big sagebrush- dominated riparian corridors to dry meadows. Two ecological site types were identified that occur naturally in the drainages and that have similar soils and landforms to serve as models for the restoration effort. The ecological site types are: (1) the dry meadow ecological site type which is dominated by grasses and sedges and has relatively shallow water tables; and (2) the basin big sagebrush/basin wild rye (Leymus cinereus Scribner & Merr.) type. This type is dominated by basin big sagebrush and basin wild rye with lesser amounts of dry meadow species and has relatively deep water tables (Weixelman et al. 1996). A restoration treatment was applied to sites with relatively shallow and deep water tables that included a prescribed burn to remove basin big sagebrush followed by seeding herbaceous species typical of both ecological site types.

In this component of the larger restoration study, we examined the effects of both water table depth and prescribed burning on the soil and plant nutrient status of basin big sagebrush-dominated riparian corridors. The effects of fire on soil nutrient status in sagebrush steppe ecosystems are generally known (Blank et al. 1994a), but fires's effects on riparian corridors or interactions with water table depth have yet to be addressed. Moreover, there is limited data on how plant nutrient content is influenced when growing in post-fire soil. Therefore, this study asks 3 questions: 1) Does soil nutrient status dif-

fer between sites with different water tables or in response to burning? 2) Does microsite influence post-fire soil nutrient status and nutrient uptake? 3) Do plants acquire soil nutrients at different rates in burned and unburned plots?

#### **Materials and Methods**

### **Study Sites**

In July 1996, we located 4 basin big sagebrush-dominated study sites along Willow (39°33'12"N, 116°59'30"W), Ledbetter (39°45'30"N, 117°30'W), and Marysville (39°02'30"N, 117°24'27"W) Creeks in the Toiyabe Mountain Range of central Nevada that represented progressive stages of modification of the dry meadow ecosystem type. Study sites were on drainageways and stream terraces at elevations ranging from 2,000 to 2,300 m, and were characterized by incised stream reaches. The climate is semiarid, and on average 72% of the 30 to 40 cm of average annual precipitation falls from Mid-October through May. Average temperature ranges from -4°C in January to 20°C in July. Soil parent material is Quaternary alluvium of mixed mineralogy. The Willow Creek drainage is derived from granite, chert, shale, quartzite, limy sandstone and siltstone, and limestone. The Marysville and Ledbetter drainages are derived from quartz latite with a lesser amount of andesite and basalt.

Site selection was based on presence of species typical of the dry meadow and basin big sagebrush/giant wild rye ecosystem types (Weixelman et al. 1996), and depths to water table determined from hand-augered wells. Two hydro-geomorphic positions were included. Two study sites were located on lower benches and were characterized by relatively shallow water tables (Willow wet and Ledbetter Creeks; -153 to -267 cm below the surface). An additional 2 sites were located on higher benches and had deeper water tables (Willow dry and Marysville Creeks; -268 to  $\geq -300$  cm below the surface). Sites with shallow water tables had higher cover of perennial graminoids and forbs, lower densities and volumes of basin big sagebrush, and were more similar to the model dry meadow type than shallow water table sites (Linnerooth et al. 1998). Graminoid species which typify the sites included Nevada bluegrass (Poa secunda spp. juncifolia [Scribner] R. Soreng), mat muhly (Muhlenbergia richardsonis [Trin] Rydb.), creeping wild rye (Leymus triticoides [Buckley] Pilger), basin wild rye,

and Douglas sedge (Carex douglassi Boott). Common forbs included lupines (Lupinus sp.) and asters (Aster sp.). The shrubs, basin big sagebrush and green rabbitbrush (Chrysothamnus viscidiflorus ssp. viscidiflorus [Hook.] Nutt.), occurred with high constancy indicating that the study sites were relatively drier than the model dry meadow ecosystem type.

Holocene pedogenesis of Quaternaryaged sediments under herbaceous vegetation has produced deep, highly organic and relatively undifferentiated profiles. The soils on all sites are classified as Pachic Haplocryolls with < 2-mm textures of loam and silt loam. Willow Creek dry and the Marysville site occur on slightly higher benches and are characterized by deeper water tables, the soils have lighter-colored mollic epipedons. Coarse fragment content for all soils lies between 5 and 50%.

# **Experimental Design and Data** Collection

A paired plot approach was used in which 1 plot at each of the 4 sites received the restoration treatment (prescribed burn and seeding), and the other plot served as a control. Burning was conducted on 19-21 Oct. 1996 to remove both woody and understory vegetation. Plots were between 740 and 900 m<sup>2</sup> in size, and fencing was used to reduce grazing and trampling by livestock and wildlife. Soils were sampled from 29 Oct. to 4 Nov. 1996. Sampling included both subcanopy and interspace microsites because of the effects of sagebrush plants on soil characteristics in these ecosystems. At each site, soil samples (n = 4) were collected for all restoration treatment/microsite combinations, and included the surface litter for unburned plots, surface ash for burned plots, and mineral soil depths of 0-2 cm, 2-5 cm, and 10-20 cm (2 water tables x 2 replicate sites x 2 burn treatments x 2 microsites x 4 depths x 4 subsamples). Surface samples were obtained with a trowel; deeper samples with a 10 cm diameter soil auger. From the time of the prescribed burn to when soil sampling occurred, approximately 12 cm of snow fell on the Willow Creek shallow and deep water table sites. Cold temperatures, however, limited snow melt and only the ash/litter layers and the upper 2 cm of the mineral soil became wet.

Vegetation was sampled in late June 1999 for both treatment/microsite combinations on the Willow Creek site with the deep water table. Three replicate transects were established within both the burned

and unburned plots, 6 or 7 sample points were randomly located along each transect (n = 20), and sample quadrats  $(0.25 \text{ m}^2)$ were placed in both the subcanopy and interspace microsites closest to the random point (2 burn treatments x 2 microsites x 3 replicate transects x 6 or 7 subsamples). All plant material was clipped at the soil surface and separated by species. The vegetation samples were dried at 60° C for 48 hours and weighed (biomass is expressed as g m<sup>-2</sup>). Only silvery lupine (Lupinus argenteus [Pursh]) and Douglas sedge on the Willow Creek site with the deep water table had sufficient biomass within all treatment/microsite combinations for analyses. This experimental design limits inferences of the results only to this site (see Hurlbert 1984).

## **Laboratory Methods**

Soil samples were stored in an ice cooler in the field and at 2°C between all preparatory steps in the lab. All soil, ash, and litter material was passed through a 2mm sieve prior to analysis. Some samples were dried slightly to facilitate sieving. All analyses were done on the < 2-mm fraction. The dichromate oxidation procedure with supplemental heating was used to quantify organic carbon (OC) (Nelson and Sommers 1982). Total N was assayed by the micro-Kjeldahl procedure (Isaac and Johnson 1976). Aqueous soil extracts were prepared by shaking 10 grams soil with 20 ml of deionized water for 1 hour (high organic matter samples required more water). The samples were centrifuged at 9400 x g for 10 minutes. Chloride, HPO<sub>4</sub>-2, and SO<sub>4</sub>-2 were quantified by gradient elution ion chromatography. Ion chromatography was used to quantify Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>+2</sup> and Ca<sup>+2</sup>. Ammonium and NO<sub>3</sub> were extracted with 0.5 M KCl (10g soil 30 ml extractant with shaking for 30 min). Ammonium was quantified using flow-injection membrane diffusion colorimetric methodology. Nitrate was quantified by ion chromatography with spectroscopic detection at 210 nm. Standards were made up from reagent grade chemicals. Soil enzyme bioassays of asparaginase, urease, and acid-phosphatase followed standard procedures (Tabatabai 1994). All data were corrected to 105°C soil weight.

Individual vegetation samples were ground in a Udy Mill and homogenized. Dry ashing followed by addition of HCl and HNO<sub>3</sub> was used to solubilize metals (Campbell and Plank 1998). Calcium, Mg, Fe, Cu, Mn, and Zn were quantified by atomic absorption spectroscopy; Na and K

were quantified by atomic emission spectroscopy; P was quantified colorimetrically using molybdenum-blue chemistry. Nitrogen in ground plant tissue was quantified by the micro-Kjeldahl method (Isaac and Johnson 1976). We used the method of Kowalenko and Van Laerhoven (1998) to solubilize total plant S which was quantified using ion chromatography.

#### **Statistical Analyses**

The overall study design was a split, split, split plot that was completely randomized at the main plot level. Water table depth (shallow, deep) was the main plot factor, treatment (burned, unburned) was the split plot factor, microsite (subcanopy, interspace) was the split, split plot factor, and depth was treated as a split, split, split plot factor (ash/litter, 0-2 cm, 2-5 cm, 10-20 cm). Sites within water tables were the main plot error term. For significant F values in the ANOVA model, Duncan's tests, at the  $\leq 0.05$  probability level of significance, were used to separate means (SAS 1996). All significant main effects and interactions were examined, but only significant treatment main effects or interactions are reported. Vegetation data were analyzed separately by species (silvery lupine and Douglas sedge) using a 2-way ANOVA. Factors were treatment and microsite (subcanopy, interspace). For significant F values, treatment differences were discerned at the  $P \le 0.05$  level using L.S. means (SAS 1996).

#### **Results and Discussion**

# Visible Effects of Prescribed Burning

Following prescribed burning, the pattern of fire effects mirrored the heterogeneous distribution of plant materials. Shrub subcanopies were covered with 1 to 5 cm of grayish to whitish ash which suggests temperatures greater than 750°C at the soil surface with lethal penetration of heat through the surface soil seedbank (Bentley and Fenner 1958). Lighter fuel loads in interspaces contributed to a darker ash generally less than 1 cm thick, which suggests temperatures somewhat greater than 350° C at this soil surface with lethal levels of heat only reaching to approximately 2 cm (Bentley and Fenner 1958). There were no visible differences in surface fire effect patterns between shallow water table and deep water table sites, but shallow water table sites had lower soil temperatures than deep water table sites (Chambers and Linnerooth 2001).

## **Influence of Water Table Depth**

The response of the shallow and deep water table sites to the burn was more a function of the vegetation composition and structure than of the hydrologic regime. Depth to water table was not a significant main effect for any of the measured attributes (Table 1). Shallow water table sites had significantly more organic carbon (OC) in undershrub microsites than interspace microsites following the burn (water table x microsite interaction; Table 1) (Table 2), whereas deep water table sites had no significant differences in OC content between microsites. The different response was the result of a greater proportional loss of OC, due to burning, in subcanopy microsites of deep rather than shallow water table sites. Similarly, prescribed burning reduced total soil N, averaged over all soil depths, but significantly more N was lost on deep rather than shallow water table sites (water table x treatment interaction; Table 1) (Table 2). Because the shallow water table sites support greater herbaceous plant cover and biomass and are generally moister than the deep water table sites (Wright 2001, Chambers and Linnerooth 2001), we expected them to contain more total soil C and N. This was not the case. For unburned sites, there were no differences in soil C and N between deep and shallow water table sites for any depth. The relatively greater loss of C and N from subcanopy microsites due to burning on deep rather than shallow water table sites can be attributed to greater fuel loads of sagebrush on deep water table sites (Linnerooth et al. 1998).

In the 0-2 cm depth increment, burning released significantly more Ca<sup>+2</sup> from deep rather than shallow water table sites (water table depth x treatment x soil depth interaction; Table 1) (Table 3). For shallow water table sites, burning released much higher levels of Mg+2 in interspace microsites than in corresponding deep water sites (water table depth x treatment x microsite x soil depth interaction; Table 1) (Table 3). Increases in aqueous-soluble cations in upper soil layers and ash are often reported following fires (Smith 1970, Marion et al. 1991, Blank and Zamudio 1998). Greater shrub fuel loads on deep water table sites may have fostered a proportionally greater increase in aqueous-soluble Ca+2 and Mg+2 due to burning than occurred on shallow water table sites. For the shallow water table sites, large increases in post-burn levels of Mg<sup>+2</sup> in ash layers of interspace microsites are possibly due to transfer of ash from subcanopy microsites.

Table 1. ANOVA results for measured soil attributes showing f-values and significance levels with an asterisk.

					Measured a	ttributes			
	df	OC1	N	C:N	pН	SO <sub>4</sub> -2	Ortho-P	Asp	Urease
$\overline{\mathrm{WT}^2}$	1	0.78	0.09	0.09	1.86	0.03	0.02	0.24	0.01
S(WT)-Err A	2								
TR	1	156.46**	352.52**	21.59*	83.58*	437.00**	11.00	35.37*	81.77*
WTxTR	1	2.24	20.73*	0.35	0.37	0.08	0.26	4.11	0.00
SxTR(WT)-ErrB	2								
MS	1	11.16*	0.44	0.03	0.95	0.81	39.57**	0.01	0.12
WTxMS	1	11.38*	1.27	1.16	0.13	0.20	6.02	0.70	0.15
TRxMS	1	38.05**	5.51	0.70	0.00	0.43	23.19**	0.39	0.11
WTxTRxMS	1	5.94	0.00	1.59	0.59	0.35	2.36	0.10	0.36
SxTRxMS(WT)-Err C	4								
D	3	61.02***	51.50***	4.16*	30.50***	7.83**	19.81***	9.24***	16.69***
WTxD	3	0.27	0.41	0.10	1.74	0.65	0.36	2.42	0.00
TRxD	3	13.30***	8.25***	0.87	34.79***	5.50**	17.59***	8.54***	20.42***
WTxTRxD	3	0.70	0.71	0.17	0.19	0.48	1.30	1.32	0.06
MSxD	3	2.36	1.99	0.38	3.64*	1.60	4.08	1.11	0.30
WTxMSxD	3	0.54	0.30	0.36	0.36	0.13	2.71	0.09	0.10
TRxMSxD	3	5.23**	2.58	0.43	2.22	0.98	5.66**	0.96	0.11
WTxTRxMSxD	3	1.09	0.06	1.70	0.58	0.39	0.99	0.02	0.09
SxTRxMSxD(WT)-Err D	22								
( )					Measured	attributes			
	df	Ptase	NO <sub>3</sub>	NH <sub>4</sub> <sup>+</sup>	K <sup>+</sup>	Mg <sup>+2</sup>	Ca <sup>+2</sup>	Na <sup>+</sup>	SAR
$WT^2$	1	1.75	0.40	2.10	0.28	0.01	0.59	0.04	0.50
S(WT)-Err A	2								
TR	1	42.86*	4.46	0.80	2.93	36.71*	100.26**	22.18*	19.8*
WTxTR	i	0.93	1.02	0.14	0.23	2.64	6.19	0.33	0.95
SxTR(WT)-ErrB	2	0.55	<u>-</u>						
MS	1	0.48	0.65	0.43	1.88	0.29	5.42	2.02	3.50
WTxMS	ĺ	0.64	0.72	10.81*	0.30	0.70	0.17	0.36	0.83
TRxMS	1	1.75	0.00	0.28	0.82	0.08	0.49	0.65	0.76
WTxTRxMS	1	4.80	0.44	0.04	0.55	2.03	1.67	1.11	1.16
SxTRxMS(WT)-Err C	4	.,							
D	3	40.37***	5.76**	8.42***	2.46	28.55***	34.37***	14.78***	8.03***
WTxD	3	1.92	0.48	0.18	0.12	2.11	8.27***	0.37	0.44
TRxD	3	55.48***	4.20*	7.95***	1.23	7.81**	12.71***	5.42**	4.21*
WTxTRxD	3	0.98	0.50	0.41	0.22	4.48*	5.54**	0.24	0.41
MSxD	3	0.31	2.30	0.33	1.43	0.16	3.56*	1.92	2.30
WTxMSxD	3	0.30	0.24	3.14*	0.15	0.56	1.30	0.50	1.06
TRxMSxD	3	1.93	1.97	0.49	1.02	0.11	4.55*	1.02	1.51
WTxTRxMSxD	3	0.64	0.23	0.36	0.47	3.90*	1.10	2.13	1.77
SxTRxMSxD(WT)-Err D	22	0.0	0.20	0.00	· · · ·				

Abbreviations:  $^{'}OC$  = organic carbon; N = total Kjeldahl nitrogen; C:N=carbon to nitrogen ratio;  $SO4^{^{'2}}$ =sulfate; Ortho-P=ortho-phosphate; Asp=asparaginase; Ptase = acid phosphatase; SAR=sodium adsorption ratio.  $^{^{'2}}WT$ =water table depth (shallow, deep); S = site (Willow wet, Willow dry, Ledbetter, and Marysville); TR=treatment (burned, control); MS=microsite (subcanopy, interspace); D = depth; \* denotes p < 0.05; \*\*\* denotes p < 0.01; \*\*\* denotes p < 0.001.

Table 2. Soil organic C, Kjeldahl N, and C:N ratios by site water table depth, microsite, soil depth, and treatment<sup>1</sup>.

		Shallow w	ater table			Deep water table					
	Subca	anopy	Inter	space	Subc	anopy	Inter	space			
Depth	Burned	Unburned	Burned	Unburned	Burned	Unburned	Burned	Unburned			
	_	_		Organic-C	(g kg <sup>-1</sup> )						
ash/litter	106±16	236±36	90±16	140±29	43±13	271±25	136±18	156±9			
0-2 cm	47±5	83±13	39±6	60±16	58±13	85±9	64±8	93±14			
2-5 cm	35±4	51±7	33±6	46±9	35±3	55±6	56±12	63±7			
10-20 cm	17±2	22±4	17±3	23±4	19±1	22±1	20±1	21±2			
				Kjeldahl-	N (g kg <sup>-1</sup> )						
ash/litter	$7.2 \pm 0.9$	15.7±1.9	9.4±1.5	8.6±1.8	5.2±1.0	17.0±1.9	$8.6 \pm 2.3$	12.9±1.7			
0-5 cm	4.6±0.5	$7.0 \pm 1.3$	$4.6 \pm 1.2$	5.6±1.4	$4.7 \pm 0.5$	$7.5 \pm 1.2$	$7.2 \pm 1.6$	$8.9 \pm 1.8$			
5-10 cm	$2.7 \pm 0.3$	$3.4 \pm 0.5$	$2.9 \pm 0.6$	$3.2 \pm 0.7$	$2.8 \pm 0.3$	4.1±0.5	4.5±1.1	$4.6 \pm 0.8$			
20-30 cm	$1.4 \pm 0.2$	1.5±0.2	$1.4 \pm 0.3$	1.8±0.3	$1.6 \pm 0.2$	$1.7 \pm 0.1$	$1.5 \pm 0.1$	$1.7 \pm 0.2$			
				C:	N						
ash/litter	13.8±1.9	15.2±1.3	9.7±1.0	16.4±1.3	7.8±1.0	16.8±2.0	14.0±1.3	12.4±2.0			
0-5 cm	10.3±0.5	12.3±0.3	$10.4 \pm 1.4$	11.2±0.6	12.1±2.4	11.8±0.6	$9.2 \pm 1.0$	11.3±1.0			
5-10 cm	13.0±0.6	15.4±0.5	$12.2 \pm 0.7$	14.8±0.5	12.1±1.0	13.9±1.0	13.6±1.1	14.5±0.9			
20-30 cm	12.8±0.7	14.3±1.3	12.3±0.6	13.2±0.3	12.2±1.4	13.0±0.3	13.2±0.5	12.8±0.5			

<sup>&</sup>lt;sup>1</sup>Variation in data shown by standard errors.

Table 3. Aqueous-soluble cations and KCl-extractable ammonium by site water table depth, microsite, soil depth, and treatment<sup>1</sup>.

		Shallow w	ater table		Deep water table					
	Subca	anopy	Inter	space	Subc	anopy	Inter	space		
Depth	Burned	Unburned	Burned	Unburned	Burned	Unburned	Burned	Unburned		
				Organic-Ca <sup>+2</sup>	(mmol kg <sup>-1</sup> )					
ash/litter	$3.4 \pm 0.3$	$4.3 \pm 0.7$	$4.1 \pm 1.0$	2.2±0.8	2.3±0.7	$3.2 \pm 0.5$	$2.6 \pm 0.7$	$3.2 \pm 0.8$		
0-2 cm	$2.9 \pm 1.0$	1.1±0.3	$1.3 \pm 0.4$	$0.9 \pm 0.1$	$7.4 \pm 1.6$	$1.4 \pm 0.3$	$3.1 \pm 0.9$	$1.3 \pm 0.4$		
2-5 cm	$1.6 \pm 0.4$	$0.8 \pm 0.1$	$1.5 \pm 0.4$	$0.5 \pm 0.1$	$2.6 \pm 0.4$	$1.6 \pm 0.7$	1.8±0.5	$1.2 \pm 0.2$		
10-20 cm	$0.6 \pm 0.1$	$0.6 \pm 0.1$	$0.5\pm0.2$	$0.4 \pm 0.1$	$0.5 \pm 0.1$	$0.4 \pm 0.1$	$0.6\pm0.2$	$0.4 \pm 0.1$		
				Mg <sup>+2</sup> (mr	nol kg <sup>-1</sup> )					
ash/litter	8.0±1.9	2.8±0.5	$13.0 \pm 5.0$	1.1±0.4	7.7±1.1	2.3±0.3	$2.7 \pm 1.4$	4.8±3.6		
0-2 cm	$2.0\pm0.5$	$0.7 \pm 0.1$	$1.8 \pm 0.9$	$0.5 \pm 0.1$	4.6±0.7	$1.0\pm0.2$	$2.2 \pm 0.7$	$0.7 \pm 0.1$		
2-5 cm	$1.2 \pm 0.3$	$0.4 \pm 0.1$	$1.0\pm0.3$	$0.3\pm0.1$	$2.3 \pm 0.5$	1.1±0.5	1.4±0.6	$0.6 \pm 0.1$		
10-20 cm	$0.3\pm0.1$	$0.2\pm0.1$	$0.3\pm0.1$	$0.2\pm0.1$	$0.4 \pm 0.1$	$0.3 \pm 0.1$	$0.3\pm0.1$	$0.2 \pm 0.1$		
				Na <sup>+</sup> ( mm	ol kg <sup>-1</sup> )					
ash/litter	$4.0\pm0.9$	1.8±0.5	$4.0 \pm 1.4$	0.8±0.2	5.8±1.0	$0.9 \pm 0.2$	1.3±0.5	1.3±0.8		
0-2 cm	$1.3 \pm 0.3$	$0.5 \pm 0.1$	$0.7 \pm 0.2$	$0.3 \pm 0.1$	$2.2\pm0.5$	$0.4 \pm 0.1$	1.5±0.4	$0.3 \pm 0.1$		
2-5 cm	$0.7 \pm 0.1$	$0.4 \pm 0.1$	$0.7 \pm 0.1$	$0.3 \pm 0.1$	$0.8 \pm 0.1$	$0.5\pm0.2$	$0.6 \pm 0.2$	$0.3 \pm 0.1$		
10-20 cm	$0.4 \pm 0.1$	$0.4 \pm 0.1$	$0.4 \pm 0.1$	$0.4 \pm 0.1$	$0.5 \pm 0.1$	$0.4 \pm 0.1$	$0.3\pm0.1$	$0.2 \pm 0.1$		
				K <sup>+</sup> (mma	ol kg <sup>-1</sup> )					
ash/litter	112±38	40.3±13.6	46.4±8.0		241±95	$17.2 \pm 3.1$	8.1±2.9	16.6±12.4		
0-2 cm	13.5±3.0	$7.0 \pm 1.7$	15.3±5.6	2.2±0.5	16.5±4.2	9.1±3.4	9.3±2.9	$1.7 \pm 0.4$		
2-5 cm	10.1±1.9	4.7±1.2	13.7±5.2	1.2±0.1	14.4±1.8	$10.0\pm4.7$	9.7±4.9	1.6±0.2		
10-20 cm	$2.3 \pm 0.5$	$2.2 \pm 0.7$	$3.3\pm1.2$	$1.9 \pm 0.6$	$4.0\pm2.0$	$2.7 \pm 0.8$	$1.9 \pm 0.8$	$1.0\pm0.2$		
				KC1-NH <sub>4</sub> + (	mmol kg <sup>-1</sup> )					
ash/litter	$2.5 \pm 1.1$	$10.9 \pm 8.4$	1.1±0.5	4.6±1.9	$1.6 \pm 0.3$	$3.8 \pm 1.0$	8.3±4.3	$10.4 \pm 1.8$		
0-2 cm	$8.4 \pm 2.1$	$0.6 \pm 0.1$	$3.6 \pm 0.9$	$0.7 \pm 0.3$	$7.5 \pm 0.8$	1.5±0.7	7.1±1.2	$2.4 \pm 0.5$		
2-5 cm	4.3±1.1	$0.7 \pm 0.3$	$3.1 \pm 1.0$	$0.3 \pm 0.1$	5.2±1.1	$1.3 \pm 0.7$	3.3±1.6	$0.9 \pm 0.3$		
10-20 cm	$0.3 \pm 0.1$	$0.1 \pm 0.1$	$0.6 \pm 0.3$	$0.1 \pm 0.1$	$0.6 \pm 0.2$	$0.1 \pm 0.1$	$0.5 \pm 0.3$	0.1±0.1		

<sup>&</sup>lt;sup>1</sup>Variation in data shown by standard errors.

In shallow water table sites, levels of NH<sub>4</sub><sup>+</sup> tended to be lower with increasing soil depth for both microsites (water table x microsite x soil depth interaction; Table 1) (Table 3). For deep water table sites, levels of NH<sub>4</sub><sup>+</sup> declined significantly with soil depth in interspace microsites, but were nearly constant for the first 3 depths in the subcanopy microsite and then declined significantly.

# **Effects of Prescribed Burning on Soil Attributes**

Significant loss of enzyme activity due to burning was largely limited to ash layers, but acid phosphatase activity was reduced to the 5 cm soil depth (urease, asparaginase and acid phosphatase all influenced by treatment x depth interaction; Table 1) (Table 4). Although not significant, enzyme activity in burned subcanopies generally was lower compared to

unburned controls. Soil enzyme activities declined due to heat deactivation (Saa et al. 1993, Hernandez et al. 1997, Staddon et al. 1998). The insulating properties of soil and brief exposure of the soil surface to high temperatures generally limits the depth of deactivation to 2 cm (Aston and Gill 1976). Greater loss of acid phosphatase activity deeper in the soil due to burning, relative to urease and asparaginase, may have occurred because acid

Table 4. Soil enzyme activities of urease, asparaginase, and acid phosphatase by site water table depth, microsite, soil depth, and treatment<sup>1</sup>.

		Shallow w	ater table		Deep water table						
	Subca	anopy	Inter	rspace	Subca	anopy	Interspace				
Depth	Burned	Unburned	Burned	Unburned	Burned	Unburned	Burned	Unburned			
				Urease (μm	ol g <sup>-1</sup> hr <sup>-1</sup> )						
ash/litter	$1.7 \pm 0.9$	36.4±7.6	$1.2 \pm 0.5$	45.8±23.3	0.9±0.2	41.5±6.9	3.8±1.9	34.6±7.8			
0-2 cm	$2.6 \pm 1.3$	6.6±0.9	$4.9 \pm 1.1$	9.9±1.6	$1.3 \pm 0.4$	9.3±1.0	$3.9 \pm 1.1$	$8.8 \pm 1.8$			
2-5 cm	$1.4 \pm 0.5$	$3.5 \pm 0.7$	$3.8 \pm 1.1$	8.2±0.9	$1.8 \pm 0.6$	$4.4 \pm 0.6$	4.4±1.2	5.3±1.1			
20 cm	5.0±0.8	$6.6 \pm 0.7$	4.0±0.9	$6.2 \pm 1.0$	$4.2 \pm 0.7$	5.6±0.7	$4.6 \pm 0.7$	5.7±0.9			
				Asparaginase (	μmol g <sup>-1</sup> hr <sup>-1</sup> )						
ash/litter	$0.9 \pm 0.4$	10.7±6.6	$0.6 \pm 0.2$	8.2±2.0	0.8±0.1	5.7±1.7	$1.0 \pm 0.5$	3.4±1.5			
0-2 cm	$0.0 \pm 0.9$	$1.0 \pm 0.1$	$1.3 \pm 0.8$	$0.8 \pm 0.2$	$0.7 \pm 0.4$	1.1±0.2	2.6±1.3	1.6±0.5			
2-5 cm	$0.8 \pm 0.2$	$0.4 \pm 0.2$	$0.0\pm0.3$	$0.9 \pm 0.1$	$0.7 \pm 0.3$	$0.4 \pm 0.2$	$0.7 \pm 0.3$	2.5±2.0			
10-20 cm	$0.2\pm0.1$	$0.7\pm0.3$	$0.1 \pm 0.2$	$0.8 \pm 0.2$	$0.3\pm0.1$	$0.3 \pm 0.1$	$0.6\pm0.3$	$0.7 \pm 0.2$			
				Acid Phosphatas	e (umol g <sup>-1</sup> hr <sup>-1</sup> )						
ash/litter	$0.1 \pm 0.1$	9.0±1.4	$0.1 \pm 0.1$	$10.4 \pm 3.3$	0.1±0.1	13.3±1.0	$0.9\pm0.8$	11.7±1.1			
0-2 cm	$1.0 \pm 0.3$	6.1±0.8	$2.0\pm0.8$	5.5±0.8	$0.7\pm0.3$	8.4±0.7	2.2±0.7	6.2±1.0			
2-5 cm	2.8±0.9	4.2±0.4	2.3±0.7	5.8±1.0	1.0±0.3	6.0±0.8	2.9±0.6	4.9±0.7			
10-20 cm	$1.6 \pm 0.2$	1.6±0.2	1.3±0.2	$2.6 \pm 0.6$	1.1±0.1	2.1±0.4	1.3±0.2	2.2±0.4			

Variation in data shown by standard errors.

Table 5. Soil pH and water-soluble sulfate and ortho-P and KCl-extractable nitrate by site water table depth, microsite, soil depth, and treatment 1.

		Shallow v	vater table		Deep water table					
	Subca	nopy	Inters	space	Subca	nopy	Inters	space		
Depth	Burned	Unburned	Burned	Unburned	Burned	Unburned	Burned	Unburned		
					Н					
ash/litter	$8.6 \pm 0.2$	$6.3 \pm 0.3$	$8.8 \pm 0.3$	5.8±0.2	9.2±0.3	$5.9 \pm 0.1$	$7.9 \pm 0.4$	5.6±0.2		
0-2 cm	$7.1 \pm 0.1$	$5.9 \pm 0.1$	$7.4 \pm 0.4$	5.8±0.2	$6.8 \pm 0.3$	$5.4 \pm 0.1$	$6.9 \pm 0.3$	5.3±0.2		
cm	$6.4 \pm 0.2$	$6.0 \pm 0.2$	$6.8 \pm 0.4$	$5.7 \pm 0.1$	$6.3 \pm 0.1$	$5.5 \pm 0.1$	$6.0\pm0.3$	5.1±0.1		
10-20 cm	$6.7 \pm 0.1$	$7.1 \pm 0.2$	$6.6 \pm 0.3$	6.5±0.2	$6.9 \pm 0.2$	$6.5 \pm 0.1$	$6.4 \pm 0.2$	$6.5 \pm 0.1$		
				Sodium ads	orption ratio					
ash/litter	1.12±0.19	$0.65 \pm 0.12$	$0.94 \pm 0.17$	$0.47 \pm 0.13$	1.83±0.29	$0.39 \pm 0.06$	$0.54 \pm 0.15$	$0.37 \pm 0.12$		
0-2 cm	$0.58 \pm 0.06$	$0.41 \pm 0.06$	$0.43 \pm 0.05$	0.28±0.03	0.62±0.11	$0.28 \pm 0.05$	$0.60\pm0.09$	$0.24 \pm 0.03$		
2-5 cm	$0.46 \pm 0.04$	$0.38 \pm 0.08$	0.45±0.05	$0.35 \pm 0.03$	$0.39 \pm 0.02$	$0.28 \pm 0.05$	$0.36 \pm 0.04$	$0.20\pm0.03$		
10-20 cm	$0.56\pm0.09$	$0.51 \pm 0.05$	$0.63\pm0.18$	$0.53\pm0.17$	0.53±0.07	$0.44 \pm 0.07$	$0.33 \pm 0.05$	$0.33 \pm 0.06$		
				SO <sub>4</sub> -2 (m	mol kg <sup>-1</sup> )					
ash/litter	17.40±5.01	$2.70\pm0.98$	14.96±5.17	0.43±0.24	17.32±3.26	1.28±0.27	5.02±1.24	$0.65 \pm 0.27$		
0-2 cm	5.62±1.47	$0.28 \pm 0.06$	4.11±1.63	$0.15 \pm 0.03$	9.40±2.46	$0.48 \pm 0.22$	5.41±1.78	$0.27 \pm 0.08$		
2-5 cm	2.73±0.86	$0.14 \pm 0.03$	3.65±1.30	$0.12 \pm 0.03$	4.17±0.94	$0.48 \pm 0.33$	$3.84 \pm 1.74$	$0.20\pm0.04$		
10-20 cm	$0.28 \pm 0.11$	$0.35 \pm 0.28$	$0.85 \pm 0.39$	$0.13\pm0.07$	1.52±0.54	$0.17 \pm 0.08$	0.97±0.51	0.28±0.16		
				ORTHO-P	(mmol kg <sup>-1</sup> )					
ash/litter	$0.26 \pm 0.11$	$2.77 \pm 0.77$	$0.18 \pm 0.05$	$0.65\pm0.26$	0.25±0.06	1.64±0.55	$0.49 \pm 0.19$	$1.10 \pm 0.45$		
0-2 cm	$0.31 \pm 0.04$	$0.65 \pm 0.12$	0.22±0.06	$0.14 \pm 0.01$	0.31±0.07	$0.66 \pm 0.13$	$0.21 \pm 0.04$	$0.20\pm0.04$		
2-5 cm	$0.30\pm0.05$	$0.59 \pm 0.09$	0.17±0.04	$0.12 \pm 0.01$	0.35±0.05	$0.60 \pm 0.14$	$0.23 \pm 0.06$	$0.23 \pm 0.05$		
10-20 cm	$0.26 \pm 0.04$	$0.33 \pm 0.05$	$0.15 \pm 0.04$	$0.13\pm0.03$	$0.18 \pm 0.04$	$0.23 \pm 0.06$	$0.09 \pm 0.02$	$0.17 \pm 0.04$		
				KCl-NO <sub>2</sub>	(mmol kg <sup>-1</sup> )					
ash/litter	$0.13 \pm 0.04$	2.87±2.15	$0.10\pm0.05$	0.66±0.55	0.05±0.01	4.18±1.30	$0.53 \pm 0.28$	$3.44 \pm 2.49$		
0-2 cm	0.50±0.31	$0.54 \pm 0.13$	1.05±0.50	1.34±0.27	$0.10\pm0.07$	$0.53 \pm 0.14$	$0.59 \pm 0.26$	$3.33\pm1.53$		
2-5 cm	0.29±0.22	$0.29 \pm 0.07$	0.37±0.15	$0.52\pm0.10$	$0.04 \pm 0.01$	$0.31 \pm 0.09$	$0.23\pm0.07$	1.41±0.45		
10-20 cm	0.13±0.02	$0.09\pm0.01$	0.13±0.03	$0.23\pm0.15$	$0.05\pm0.01$	$0.19 \pm 0.07$	$0.08 \pm 0.03$	$0.16 \pm 0.03$		

Variation in data shown by standard errors.

phosphatase is more sensitive to heat, or compound(s) created by the fire may be inhibitory to acid phosphatase activity. Reduced enzyme activities of urease, asparaginase, and acid phosphatase can potentially lower mineralization rates of N and P, at least in ash layers. The large reduction of acid phosphatase enzyme activity in burned subcanopies suggests possible interactions with P nutrition for plant establishment and regrowth.

Burned plots had lower organic carbon (OC) in the ash/litter and 0-2 cm depths than controls, with the greatest loss of OC occurring in shrub subcanopy microsites (treatment x microsite x soil depth interaction; Table 1) (Table 2). Nitrogen content was lower for all depths on burned than control plots, but was significantly lower only in ash/litter layers (treatment x depth interaction; Table 1) (Table 2). Carbon to N ratios were lower on burned plots (Table 2). Vegetation combustion and concomitant soil heating often leads to system loss of C and N and the level of loss is a function of microsite fuel levels (Schnitzer and Hoffman 1964, Raison 1979, DeBell and Ralston 1970, White et al. 1973, Blank et al. 1994a, Johnson et al. 1998). A consequence of lowered C:N ratios following the burns may be greater N mineralization potential (Christensen

1973, Stock and Lewis 1986, Dyrness et al. 1989, Kutiel and Naveh 1987).

Aqueous-soluble Na+ and KClextractable NH<sub>4</sub><sup>+</sup> exhibited a significant treatment x soil depth interaction (Table 1). Prescribed burned plots had greater content of Na<sup>+</sup> in the ash/litter, 0-2 cm, and 2-5 cm depths than control plots (Table 3). Content of KCl-extractable NH<sub>4</sub><sup>+</sup> was lower in the ash layer and higher in the 0-2 cm and 2-5 cm depths of the burned than control plots (Table 3). Large quantities of NH<sub>4</sub><sup>+</sup> can be liberated after burns (Raison 1979, Khanna and Raison 1986, Overby and Perry 1996) depending on soil burn temperature (Marion et al. 1991). Extremely high temperatures, as occurred in the ash layers, fostered a slight increase in ammonium or slight loss relative to unburned litter layers. Beneath the ash layers where soil temperature was lower, ammonium levels increased markedly. The elevated extractable ammonium values in ash and mineral layers in burned treatments relative to controls suggests enhanced N nutrition for plants. High data variability and a conservative statistical approach negates any post-fire inferences regarding aqueous-soluble K<sup>+</sup>. However, K+ was consistently higher in burned than control plots for all depths except 10-20 cm (Table 3).

Soil pH, sodium adsorption ratio (SAR), aqueous-soluble  $SO_4^{-2}$ , and KCl-extractable  $NO_3^-$  exhibited significant treatment x depth interactions (Table 1). Burning resulted in an increase in soil pH, SAR, and SO<sub>4</sub><sup>-2</sup> in all depths except 10–20 cm (Table 5). Elevated soil reaction often occurs after fire and has been attributed to hydrolyses of basic cations (Sampson 1944). The pH's attained in our study due to burning, at least in ash layers, would negatively influence nutrient availability (Marschner 1995). The increase in SAR is due to a greater proportional release of Na<sup>+</sup> than Ca<sup>+2</sup> plus Mg<sup>+2</sup> due to burning. The SARs are below threshold values of about 10 which could lead to soil dispersion (Bohn et al. 1979). However, prescribed fires undertaken in soils with higher initial SARs may result in soil dispersion. Of all attributes measured, aqueous soluble  $SO_4^{-2}$  increased the most following the burn, and was over 10 times greater in ash and mineral layers to 5 cm in burned than control plots. Although S readily volatilizes from vegetation heated above 300° C (Tiedemann 1987), oxidation of S-containing functional groups, possibly aided by catalysis on soil mineral surfaces, results in elevated soil sulfate levels after fires (Blank et al. 1994a). The large increase of aqueous-soluble SO<sub>4</sub>-2

due to fire should favorably affect S nutrition of plants as the macronutrient is assimilated dominantly by root uptake (Marschner 1995).

In subcanopy microsites, ash layers had less ortho-P than the litter layer (treatment x microsite x depth interaction; Table 1) (Table 5). Soil content of ortho-P following vegetation fires is difficult to predict due to its involvement in numerous sorption and precipitation reactions, and to the interaction of these reactions with factors such as soil type, burn temperature, and potential volatilization (Raison et al. 1985). A decline in labile pools of ortho-P following fire can be due to precipitation with Fe, Al, and Ca (Smith 1970).

Burning resulted in loss of NO<sub>3</sub><sup>-</sup> relative to controls for the litter/ash and 0–2 cm depths (treatment x soil depth interaction; Table 1) (Table 5). Loss of nitrate in surface soil layers following wildfire is almost universally reported in the literature and likely results from conversion to nitrogen gas and/or gaseous oxides of nitrogen at temperatures greater than 200° C (DeBell and Ralston 1970, White et al. 1973). There may not be much ecological significance to these NO<sub>3</sub><sup>-</sup> losses in sagebrush communities where mineralization can rapidly replenish NO<sub>3</sub><sup>-</sup> (Blank et al. 1994a).

# **Biomass and Nutrient Content of Vegetation**

In 1999, three years after the prescribed burn, total above ground herbaceous biomass for both the shallow and deep water table sites averaged 1,243 g m<sup>-2</sup> on the burned plots and 469 g m<sup>-2</sup> on control plots (Wright 2001). At the Willow Creek site with the shallow water table, the nitrogen fixing legume, silvery lupine, had higher aboveground biomass on the burned than control plot (Table 6). Aboveground biomass of Douglas sedge was higher on burned interspaces, but lower on burned subcanopies than corresponding controls. Tissue nutrient content of the 2 species varied considerably by treatment and microsite (Table 6). Silvery lupine had higher tissue content of N and lower content of Ca on the burned than control plot. Also, tissue content of Zn and Fe was higher on subcanopies than interspaces of the burned plot. For Douglas sedge, only tissue S was higher on the burned than control plot. Tissue Zn and K of Douglas sedge were higher in plants growing on the control than burned plot, and tissue P was higher in subcanopies than interspaces on the control.

The stimulation of plant growth following fires or "ash-bed effect" is attributed to increased nutrient availability and uptake by plants (Cromer 1967, Kutiel and Naveh 1987). Luxury consumption of nutrients

after fires may promote growth and reproduction in low nutrient soils (Rundel and Parson 1980). The influence of the "ashbed effect" may be short-lived, lasting less than one year (Renbuss 1968, Griffen and Friedel 1984). This may explain why we did not see overwhelming evidence of elevated nutrient uptake in burned plots. In a pot study, Blank et al. (1994b) grew big sagebrush (Artemisia tridentata spp. Wyomingensis Beatle & A. Young), cheatgrass (Bromus tectorum L.), bottlebrush squirreltail (Elymus elymoides (Raf.) Swezey), and Indian ricegrass (Achnatherum hymenoides (Roemer & Schultes) Barkworth) in the surface 5 cm of burned and unburned sagebrush subcanopy soil. Bottlebrush squirreltail and cheatgrass produced significantly more aboveground biomass when grown in burned soil, yet aboveground tissue concentrations of N, K, P, Ca, and Mg (S and micronutrients were not quantified in this study) did not differ significantly between treatments. Although higher tissue concentrations of S, Fe, and Zn suggest increased uptake due to elevated soil availability on the burned plot, other factors may contribute to the increased abundance of herbaceous vegetation. Removal of sagebrush can enhance water availability (Sturges 1983), and the burned plots had higher late season moisture availability than the control plots (Chambers and

Table 6. Biomass and elemental content of aboveground plant material of Silvery lupine and Douglas sedge, by microsite and treatment, for the Willow Wet study site 1.

	Su	bcano	ру	Inter	rspac	e	Micro	osites F	Pooled
Attribute	Burned		Unburned	Burned		Unburned	Burned		Unburned
				Silvery lupine					
Biomass (g 0.25m <sup>-2</sup> )	2.21±1.35	*	1.24±0.52	9.64±3.98	*	2.67±1.34	5.19±1.07		1.61±0.73
N (mol kg <sup>-1</sup> )	$2.6 \pm 0.1$		$1.9 \pm 0.1$	$2.3 \pm 0.2$		$1.8 \pm 0.1$	$2.4 \pm 0.1$	*	$1.9 \pm 0.1$
Zn (mmol kg <sup>-1</sup> )	0.58±0.05	*	$0.37 \pm 0.02$	$0.39 \pm 0.03$	*	$0.50\pm0.03$	$0.46 \pm 0.04$		$0.69 \pm 0.03$
Mn (mmol kg <sup>-1</sup> )	$0.78 \pm 0.12$		$0.68 \pm 0.16$	$0.52 \pm 0.04$		$0.71 \pm 0.15$	$0.62 \pm 0.07$		$0.69 \pm 0.10$
Fe (mmol kg <sup>-1</sup> )	$8.8 \pm 3.4$	*	$1.7 \pm 0.1$	$2.7 \pm 0.9$		4.3±1.1	$5.0 \pm 1.7$		$3.0\pm0.7$
P (mmol kg <sup>-1</sup> )	74±6		72±1	78±4		75±5	76±3		73±3
K (mol kg <sup>-1</sup> )	$0.38 \pm 0.03$		$0.45 \pm 0.03$	$0.38 \pm 0.04$		$0.36 \pm 0.05$	$0.38 \pm 0.03$		$0.41 \pm 0.03$
Na (mmol kg <sup>-1</sup> )	9±1		19±8	17 <b>±</b> 7		33±10	14±5		26±6
Ca (mol kg <sup>-1</sup> )	$0.22 \pm 0.01$		$0.30 \pm 0.02$	$0.27 \pm 0.03$		$0.30 \pm 0.02$	$0.25 \pm 0.02$	*	$0.30 \pm 0.01$
S (mmol kg <sup>-1</sup> )	39.0±13.7		35.9±0.7	32.9±3.8		29.5±2.7	$35.2 \pm 2.6$		32.7±1.7
				Douglas sedge	2				
Biomass (g m <sup>-2</sup> )	5.53±1.61	*	1.28±0.63	0.49±2.84	*	4.81±1.76	3.36±0.79		2.21±1.35
N (mol kg <sup>-1</sup> )	$1.1 \pm 0.1$		$1.0 \pm 0.1$	$1.3 \pm 0.2$		$1.2 \pm 0.1$	$1.2 \pm 0.1$		$1.2 \pm 0.1$
Zn (mmol kg <sup>-1</sup> )	$0.41 \pm 0.02$		$0.54 \pm 0.05$	$0.35 \pm 0.02$		$0.50 \pm 0.05$	$0.37 \pm 0.02$	*	$0.51 \pm 0.03$
Mn (mmol kg <sup>-1</sup> )	$0.58 \pm 0.06$		$0.52 \pm 0.09$	$0.57 \pm 0.05$		$0.67 \pm 0.07$	$0.57 \pm 0.04$		$0.62 \pm 0.06$
Fe (mmol kg <sup>-1</sup> )	5.7±1.0		$4.6 \pm 1.0$	$3.2 \pm 0.6$		$3.0\pm0.4$	$4.3 \pm 0.7$		$3.5 \pm 0.5$
P (mmol kg <sup>-1</sup> )	53±2	*	79±3	55±4		59±1	54±2		65±3
K (mol kg <sup>-1</sup> )	$0.37 \pm 0.02$		$0.49 \pm 0.02$	$0.41 \pm 0.01$		$0.44 \pm 0.03$	$0.39 \pm 0.01$	*	$0.46 \pm 0.02$
Na (mmol kg <sup>-1</sup> )	11±2	*	69±27	13±2		24±11	12±1		39±13
Ca (mol kg <sup>-1</sup> )	$0.14 \pm 0.01$		$0.11 \pm 0.01$	$0.13 \pm 0.01$		$0.14 \pm 0.01$	$0.13 \pm 0.01$		$0.13 \pm 0.01$
S (mmol kg <sup>-1</sup> )	49.2±4.3		49.2±1.3	57.7±5.0		45.8±2.1	53.9±3.5	*	46.4±1.5

An asterisk denotes significant differences due to treatment; significance judged at the p≤0.05 level using Fisher's protected least significant difference. Variation in data shown by standard errors.

Linnerooth 2001). Shading by sagebrush can reduce nutrient uptake by herbaceous vegetation (Cui and Caldwell 1997). Kutiel and Naveh (1987) report significant increase in shoot and root biomass and elemental content of N, P, Mg, K, Ca, and Fe 2 months after fire for plants grown on ash.

### **Conclusions and Implications**

This and companion studies show that while shallow water table sites can be restored to graminoid dominance with prescribed fire, deep water table sites appear to have crossed thresholds and will require both burning and reseeding. Three years after the burns, the shallow water table sites were dominated by species that characterize the dry meadow ecological site as a result of seedling establishment (Chambers and Linnerooth 2001) and especially vegetative growth and expansion of preexisting understory species (Wright 2001). In contrast, the deep water table sites were dominated by annual weedy species that occurred in the seed bank, including lambsquarter (Chenopodium album L.) and tansy mustard (Descurainia pinnata [Walt.] Britt.) (Wright 2001. Chambers and Wehking, unpublished data).

The soil response to the burns on the shallow vs deep water table sites was only indirectly related to the water table regime. There were no detectable differences between shallow and deep water table sites for the soil attributes measured. However, deep water table sites lost proportionately more C and N from subcanopy microsites, and released more aqueous soluble Ca<sup>+2</sup> and Mg<sup>+2</sup>, than shallow water table sites. Also, higher mortality of both seeds and vegetative propagules occurred in subcanopy than interspace microsites, especially on deep water table sites (Wehking and Chambers, unpublished data). These differences can be attributed to higher fuel loads of basin big sagebrush (Linnerooth et al. 1998) and higher burn temperatures on deep rather than shallow water table sites. The water table regime influences the persistence and growth of basin big sagebrush in riparian corridors. Moreover, the biomass of basin big sagebrush, relative to understory herbaceous vegetation in degraded riparian corridors, can influence the effects of prescribed burns on both soils and vegetation. Prescribed burns on sites with high volumes of sagebrush and deep water tables can have more negative effects than on sites with lower volumes of sagebrush and shallower water tables.

The response of the soil chemical properties to prescribed burning were consistent with the existing literature, i.e., there is a general increase in water-soluble Ca<sup>+2</sup>, Mg<sup>+2</sup>, Na<sup>+</sup>, SO<sub>4</sub><sup>-2</sup>, and K<sup>+</sup>, and KCl-extractable NH<sub>4</sub><sup>+</sup> and a loss of extractable NO<sub>3</sub>. The increase in available nutrients undoubtedly influenced the response of both the perennial and herbaceous species following the burns. Even 3 years after burning, the content of selected nutrients in both silvery lupine and Douglas sedge was higher on burned than control plots on the deep water table site in Willow Creek. Our overall results indicate that prescribed burns on basin big sagebrush dominated sites with shallow water tables and a preexisting understory of dry meadow species result in minimal C and N loss and increased available nutrients for vegetative regrowth and seedling establishment. Burning sites with deep water tables, high sagebrush biomass, and few perennial herbaceous species can result in proportionately greater C and N loss, and increase available nutrients for weedy species. While burning alone is an appropriate restoration treatment for shallow water table sites, deep water table sites will require a burning prescription that minimizes fire severity, and reseeding with species adapted to more xeric conditions.

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